

DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR THIRD YEAR B.TECH

Scheme of Teaching and Examination: Semester- V (Food Technology)

Subject							Examination Scheme (Marks)						
Code		_	an a	_	G 114	Theory			Practical				
		L	T	P	Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing		
FT311	Technology of Cereals and Bakery Products	04	-	-	04	CIE SEE	50 50	20 20	-	-	-		
FT312	Fruits and Vegetables Processing Technology	04	-	-	04	CIE SEE	50 50	20 20	-	-	-		
FT313	Dairy Technology	04	-	-	04	CIE SEE	50 50	20 20	-	-	-		
FT314	Food Process Engineering-II	04	-	-	04	CIE SEE	50 50	20 20	-	-	-		
FT315	Process Instrumentation, Dynamics and control	04	-	-	04	CIE SEE	50 50	20 20	-	-	-		
FT311L	Technology of Cereals & Bakery Products Laboratory	-	-	02	01	-	-	-	EOE	50	20		
FT312L	Fruits& Vegetables Processing Technology Laboratory	-	-	02	01	-	-	-	EPE IOE	50	20 20		
FT313L	Dairy Technology Laboratory	-	-	02	01	-	-	-	EPE	50	20		
FT314L	Food Process Engineering-II Laboratory	П	-	02	01	-	-	-	IOE	50	20		
FT315L	Process Instrumentation, Dynamics and control Laboratory	-	-	02	01	-	-	-	IOE	50	20		
	Total	20	-	10	25	-	500	-	-	300	-		

Audit Course III											
LS311	Introduction to Foreign Language	02	-	-	-	Institute Level	-	-	-	-	-

Total contact hours per week: 30+2=32

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students.

CIE: Continuous Internal Evaluation SEE: Semester End Examination IPE: Internal Practical Evaluation EPE: External Practical Examination IOE: Internal Oral Evaluation EOE: External Oral Examination



DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR THIRD YEAR B.TECH

Scheme of Teaching and Examination: Semester- VI (Food Technology)

Subject	Subject	Teaching Scheme with Credits Examination Scheme (Mar (Hours / Week)					(arks)				
Code					~		Theory			Practica	l
		L	T	P	Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing
FT321	Sugar and Confectionery Technology	04	-	-	04	CIE SEE	50 50	20	-	-	-
FT322	Food Packaging	04	-	-	04	CIE SEE	50	20	-	-	-
FT323	Biochemical Engineering	04	-	-	04	CIE SEE	50	20	-	-	-
FT324	Process Equipment Design and Drawing	04	-	-	04	CIE SEE	50	20	-	-	-
FT325	Industrial Economics and Management	04	-	-	04	CIE SEE	50	20 20	-	-	-
FT321L	Sugar and Confectionery Technology Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT322L	Food Packaging Laboratory	-	-	02	01	-	-	-	EOE	50	20
FT323L	Biochemical Engineering Laboratory	-	-	02	01	-	-	-	EOE	50	20
FT324L	Process Equipment Design and Drawing Laboratory	-	-	02	01	-	-	-	IOE	50	20
FT325L	Seminar and Mini Project	-	-	02	01	-	-	-	IOE	100	40
	Total	20	-	10	25	-	500	-	-	300	-
	1			Δudi	t Course IV	7	l	I	I		I
RM321	Research Methodology	02	-	-	-	-	-	-	-	-	-

Total contact hours per week: 30+2=32

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students

CIE: Continuous Internal Evaluation SEE: Semester End Examination IPE: Internal Practical Evaluation EPE: External Practical Examination EOE: External Oral Examination

Seminar and Mini Project: (Contact hours: 2)

Seminar which will carry 50 marks will be delivered by individual students in front of teachers panel based on Seminar Report and Seminar oral presentation.

Mini Project which will carry 50 marks will be carried out by group of maximum 4 students. Mini project will be based on market survey and product or model development. Product or model will be finalized by respective guide. Report should be submitted to teachers' panel. Evaluation will be based on market survey report and project/ Model Viva-Voce.

Credit: 1

Equivalence of Third Year B.Tech (Food Technology) Semester V and VI

The above detailed syllabus is a revised version of the Third Year BTech (Food Technology) Program being conducted by the Shivaji University at the Technology Department of the University. This syllabus is to be implemented from June 2018, (Academic year 2018-19). The prime feature of this revision is the transformation of the existing curriculum into the Outcome based curriculum as specified in NBA rules and regulations.

The Equivalence for the subjects/courses of Food Technology at Third Year B Tech Semester V and VI pre-revised Program under the faculty of Engineering and Technology is as follows.

Third Year B.Tech Semester V (Food Technology)

Sr.No	Third Year B.Tech(Food Technology) Semester V Pre-revised syllabus	Third Year B.Tech(Food Technology) Semester V Revised syllabus	Remark
1.	Process Instrumentation,	Process Instrumentation,	Change in number of lectures and
	Dynamics & Control	Dynamics & Control	content
2.	Food Process Engineering-II	Food Process Engineering- II	Change in number of lectures and content
3.	Fruits and Vegetables	Fruits& Vegetables	Change in number of lectures and
	Processing Technology	Processing Technology	content
4.	Technology of Cereals &	Technology of Cereals and	Change in number of lectures and
	Bakery Products	Bakery Products	content
5.	Dairy Technology	Dairy Technology	Change in number of lectures and content
6.	D 1	Process Instrumentation,	Change in nomenclature
	Process Instrumentation,	Dynamics & Control	Change in number and titles of
	Dynamics & Control Lab	Laboratory	practicals
7.	Food Process Engineering-II	Food Process Engineering-	Change in nomenclature
	Lab	II Lab Laboratory	Change in number and titles of practicals
8.	Fruits& Vegetables Processing	Fruits& Vegetables	Change in nomenclature
	Technology Lab	Processing Technology Laboratory	Change in number and titles of practicals
9.	Tasky alogy of Canada Pr	Technology of Cereals &	Change in nomenclature
	Technology of Cereals & Bakery Products Lab	Bakery Products	Change in number and titles of
	•	Laboratory	practicals
10.	Dairy Technology Lab	Dairy Technology	Change in nomenclature
		Laboratory	Change in number and titles of practicals
11.	Presentation and	-	Shifted to Semester IV as soft skill
	Communication		development
12.	-	Introduction to Foreign	Shifted from Semester III for
		Language	uniformity with Department with change in number of lectures and
			content

Third Year B.Tech Semester VI (Food Technology)

Sr.No	Third Year B Tech(Food	Third Year B.Tech(Food	Remark
	Technology) Semester VI Pre-revised syllabus	Technology) Semester VI Revised syllabus	
1.	Sugar and Confectionery	Sugar and Confectionery	Change in number of lectures and
	Technology	Technology	content
2.	Food Packaging	Food Packaging	Change in number of lectures and content
3.	Industrial Economics and	Industrial Economics and	Change in number of lectures and
	Management	Management	content
4.	Biochemical Engineering	Biochemical Engineering	Change in number of lectures and content
5.	Process Equipment Design And Drawing	Process Equipment Design And Drawing	Change in number of lectures and content
6.	Sugar and Confectionery	Sugar and Confectionery	Change in nomenclature
	Technology Lab	Technology Laboratory	Change in number and titles of practicals
7.	Food Packaging lab	Food Packaging Laboratory	Change in number and titles of practicals
8.	Biochemical Engineering lab	Biochemical Engineering Laboratory	Change in nomenclature Change in number and titles of practicals
9.	Process Equipment Design And Drawing lab	Process Equipment Design And Drawing Laboratory	Change in number and titles of practicals
10.	Mini Project	Seminar & Mini Project	Change in nomenclature and content
11.	Audit Course III	Audit Course III	Change in number of lectures and
	Research Methodology	Research Methodology	content

Audit course have not been assigned any credits. The students will be evaluated for these courses by the concerned course in charge. There will be grade conferred to the student. The grade will be based on conversion of marks obtained out (Obtaining passing grade is essential). Please refer to chart in the detail examination scheme. The chart shows the marks range and the respective grade.



DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR FINAL YEAR B.TECH

Scheme of Teaching and Examination: Semester- VII (Food Technology)

Subject	Subject	Tea	(Credi	eme with ts Veek)		rks)					
Code	, and the second						Theory]	Practical		
		L	Т	P	Total Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing	
FT 411	Meat, Poultry and Fish Processing Technology	04	-	-	04	CIE SEE	50 50	20 20	-	-	-	
FT 412	Legume and Oilseed Technology	04	-	-	04	CIE SEE	50 50	20 20	-	-	-	
FT413	Food Quality and Safety Management	04	-	-	04	CIE SEE	50 50	20 20	-	-	-	
FT414	Food Biotechnology	04	-	-	04	CIE SEE	50 50	20 20	-	-	-	
FT415	Elective –I	03	_	-	03	CIE SEE	50 50	20 20	-	-	-	
FT416	Industrial Training Report	-	1	-	01	-	-	-	IOE	50	20	
FT411L	Meat, Poultry and Fish Processing Technology Laboratory	-	-	02	01	-	-	-	EOE	50	20	
FT412L	Legume and Oilseed Technology	-	-	02	01	-	-	-	EPE	50	20	
FT413L	Food Biotechnology		-	02	01	-	-	-	IOE	50	20	
FT414L	Major Project-Phase I	-	-	04	02	-	-		IOE	100	20	
	Total	19	1	10	25	-	500	-	-	300	-	

	Audit Course V										
HS411	Introduction to Indian Constitution	02	-	-	-	Institute Level					

Total contact hours per week: 30+2=32

Total contact hours per week. co. 2 - c2	
	Elective I
1. Functional Foods & Nutraceuticals	: (FT 415.1)
2. Beverages Technology	: (FT 415.2)
3.Refrigeration Engineering :	(FT 415.3)
4.Database Management :	(FT 415.4)

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students

CIE: Continuous Internal Evaluation SEE: Semester End Examination
IPE: Internal Practical Evaluation EPE: External Practical Examination
IOE: Internal Oral Evaluation EOE: External Oral Examination

<u>Note:</u> There will be an industrial training for minimum 30 days after semester VI. This will cover the study of a reputed Food industry from standpoint of Food Technology. The students will submit a report of the training. This particular activity is equivalent to one credit and it carries 50 marks as an Internal Oral Evaluation (IOE) which is included in Semester VII. The students will follow one specific format recommended by the Program Advisory Board for submission of their industrial training report.

MAJOR PROJECT (Phase I)

Teaching Scheme: P: 4 hour/week

The students are required to carry out one of the following projects related to field of Food Technology.

Credits: 02

- 1. Process based Project: Manufacture of product.
- 2. Equipment based Project: Detailed design and fabrication of the equipment for a given capacity.
- 3. Experiment based Project: Experimental investigation of basic or applied research problem.
- 4. Industrial Problems: Any problem or project directly related to existing plants for modification of process or equipment or regarding pollution control and energy conservation under the guidance of one or more faculty members.

The activity will be undertaken at the beginning of the seventh semester in consultation with concerned guide and it must be completed in eighth semester. The project work is to be carried out by a group of students (not more than five students in a group).

The students will submit the report to the respective guide, present their work at the end of Semester. They need to cover the following aspects related to their project topic:

- Introduction to the Project topic
- Detailed Literature Survey on the topic.
- Plan/outline of the Project work.
- Submitting requirements for execution the project work.

Based on the first phase work, it is desirable that the project group will publish a review article in a reputed and relevant Journal. The project group has to deliver Project Progress presentation using LCD provided by the institute.

The project progress will be evaluated by a research and review committee of internal teachers. The committee includes concerned guide, the Program Coordinator, and other faculty members at the Department. This particular activity is assigned 100 marks as an IOE. The committee will evaluate the performance of the project group.

The students will follow one specific format recommended by the Program Advisory Board for submission of their Project Report.

INDUSTRIAL TRAINING REPORT

Credit: 1

The industrial training after Semester VI and before commencement of Semester VII will cover the study of reputed food industries from standpoint of food technology. The report of the training is required to be submitted by the students. The evaluation of this particular activity is included in this Semester VII.



DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR **FINAL YEAR B.TECH**

Scheme of Teaching and Examination: Semester- VIII (Food Technology)

Subject	Subject			edits	ne with	Examination Scheme (Marks)					
Code							Theory		P	ractical	
		L	Т	P	Total	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing
FT421	Post Harvest Technology of Plantation Crops	04	_	_	04	CIE SEE	50 50	20 20	-	-	-
FT422	Design and Development of New Products	04	-	-	04	CIE SEE	50 50	20 20	-	-	-
FT423	Waste Management of Food Industries	03	-	_	03	CIE SEE	50 50	20 20	-	-	-
FT424	Elective-II	03	-	_	03	CIE SEE	50	20	-	-	-
FT425	Food Plant Design and Layout	04	_	_	04	CIE SEE	50	20 20	-		-
FT426	Entrepreneurship Development for Food Technologists	02	-	-	02	-	-	-	IOE	50	20
FT421L	Post Harvest Technology of Plantation Crops Laboratory	-	-	02	01	-	-	-	IOE	50	20
FT422L	Design and Development of New Products Laboratory	-	-	02	01	-	-	-	EOE	50	20
FT423L	Major Draiget Phase II	-	-	06	03	-	-	-	IOE	100	40
F1423L	Major Project-Phase II			00	03	-	-	-	EOE	50	20
	Total	20	-	10	25	-	500	-	-	300	-
		•	•	•	Audit C	ourse VI		•		•	
HS422	Professional Ethics	02	-	-	-	Institute Level	-	-	-	-	-

Total contact hours per week: 30+2=32

	Elective II		
1.	Flavors Technology	:	(FT 424.1)
2.	Snack Foods Technology	:	(FT 424.2)
3.	Sugar Technology	:	(FT 424.3)
4.	Energy Systems and Technology	:	(FT 424.4)

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students

Continuous Internal Evaluation CIE: SEE: Semester End Examination IPE: **Internal Practical Evaluation External Practical Examination** EPE: IOE: Internal Oral Evaluation **External Oral Examination** EOE:

ELECTIVE-II (OPEN ELECTIVE)

Besides the departmental electives there is a concept of open elective listed under

Teaching Scheme: L: 3 hours/week

Credits: 3

The interested students have to choose the open elective from any of the specialized program available on the campus. The students have to contact the concerned course teacher and attend the classes in the respective course which will be taught by the concerned teacher.

MAJOR PROJECT (PHASE II)

Teaching Scheme: P: 06hours/week

Credits: 3

In the second phase of the Project Work, all the project groups will carry out actual execution the work planned as stated as an outcome of the first phase in the seventh semester. They will submit the final project report in two hard bound copies along with soft copy to the guide. The guide will submit one copy to the Program office and the other copy will be owned by him/her. The report will be prepared in a standard format as provided by the Program Advisory Board.

Generally, the report will consist of Introduction, Review of Literature, Materials and Methods, Result & Discussion, Summary & Conclusion and References. There may be little variation in project report writing depending on the nature of the respective project problem under investigation.

The objective of the project is to make use of the knowledge gained by the students at various stages of the B.Tech Food Technology Program. This helps to judge the level of proficiency, originality and capacity for application of the knowledge attained by the students at the end of the studies.

This particular activity will be for 100 marks as an Internal Practical Evaluation (IPE). The same research and review committee will carry out the assessment of the project groups.

For External Practical Examination (EPE) carrying 50 marks, there will be a external examiners to assess the project work. The research and review committee along with the external examiner will examine each of the project groups through VIVA VOCE and physical verification of the project models if any. The students group using the LCD will demonstrate their work to all the examiners.

Equivalence of Final Year B.Tech (Food Technology) Semester VII and VIII

The above detailed syllabus is a revised version of the Final Year. BTech (Food Technology) course being conducted by the Shivaji University at the Technology Department of the University. This syllabus is to be implemented from June 2019, (Academic year 2019-20). The prime feature of this revision is the transformation of the existing curriculum into the Outcome based curriculum as specified in NBA rules and regulations.

The Equivalence for the subjects of Food Technology at Final Year B Tech Semester VII and VIII pre-revised course under the faculty of Engineering and Technology is as follows.

Final Year B.Tech Semester VII (Food Technology)

Sr.No	Final Year B.Tech(Food Technology) Semester VII Pre-revised syllabus	Final Year B.Tech(Food Technology) Semester VII Revised syllabus	Remark
1.	Meat, Poultry and Fish Processing Technology	Meat, Poultry and Fish Processing Technology	Change in no of lectures
2.	Legume and Oilseed Technology	Legume and Oilseed Technology	Change in no of lectures
3.	Food Quality & Safety Management	Food Quality & Safety Management	Change in no of lectures
4.	Food Biotechnology	Food Biotechnology	Change in no of lectures
5.	Elective –I	Elective –I	No change
6.	Major Project Phase I	Major Project Phase I	Change in no of lectures
7.	Laboratory- I Meat, Poultry and Fish Technology Lab	Meat, Poultry and Fish Technology Laboratory	Slight change in nomenclature
8.	Laboratory- II Legume and Oilseed Technology Lab	Legume and Oilseed Technology Laboratory	Slight change in nomenclature
9.	Laboratory- III Food Quality & Safety Management Lab	-	Removed the lab and the practical are submerged in other subjects
10.	Laboratory- IV Food Biotechnology Lab	Food Biotechnology Laboratory	Slight change in nomenclature
11.	Seminar & Industrial Training Report	Industrial Training Report	Seminar shifted to Semester VI
12.	Audit Course V Professional Ethics	-	Shifted to semester VIII
13.	-	Audit Course V Introduction to Indian Constitution	Shifted from semester VIII

Final Year B.Tech Semester VIII (Food Technology)

Sr.No	Final Year B.Tech (Food Technology) Semester VIII	Final Year B.Tech(Food Technology) Semester VIII	Remark
	Pre-revised syllabus	Revised syllabus	
1.	Post Harvest Technology of Plantation	Post Harvest Technology of	Change in no of lectures
	Crops	Plantation Crops	
2.	Design and Development of New	Design and Development of	Change in no of lectures
	Products	New Products	
3.	Waste Management of Food Industries	Waste Management of Food	No change
	Waste Management of Food madistres	Industries	
4.	Elective-II	Elective-II	No change
5.	Food Plant Design and Layout	Food Plant Design and Layout	Change in no of lectures
6.	Major Project (Phase-II)	Major Project (Phase-II)	Change in no of lectures
7.	Post Harvest Technology of Plantation	Post Harvest Technology of	Change in nomenclature
	Crops Lab	Plantation Crops Laboratory	
8.	Design and Development of New	Design and Development of	Change in nomenclature
	Products Lab	New Products laboratory	
9.	Entrepreneurship Development for	Entrepreneurship	No change
	Food Technologists	Development for Food	
		Technologists	
10.	Audit Course VI		Shifted to semester VII with title
	Introduction to Constitution of India	-	change.
11.	Waste Management of Food Industries		Removed the lab and the practical
	Lab	-	are submerged in other subjects
12.		Audit Course VI	Shifted from semester VII
		Professional Ethics	

Audit course have not been assigned any credits. The students will be evaluated for these courses by the concerned course in charge. There will be grade conferred to the student. The grade will be based on conversion of marks obtained out of 50. (Obtaining passing grade is essential). Please refer to chart in the detail examination scheme. The chart shows the marks range and the respective grade.



DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR SECOND YEAR B.TECH

Scheme of Teaching and Examination: Semester- III (Food Technology)

		Cre	Teaching Scheme with Credits (Hours / Week)			Examina	tion Sche	me (Marks	3)		
Subject	Subject					Theory			Practical		
Code	Subject	L	Т	P	Total Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing
FT211	Principles of Food Preservation	04	-	-	04	CIE SEE	50 50	20 20	-	-	-
FT212	Engineering Mathematics-III	03	1	_	04	CIE SEE	50 50	20 20	IOE -	50	20
FT213	Food Microbiology	04	_	_	04	CIE SEE	50 50	20 20	-	-	
FT214	Food Chemistry					CIE	50	20	-	-	-
	1 ood enemous	04	-	-	04	SEE	50	20	-	-	-
FT215	Heat Transfer	03	_	-	03	CIE SEE	50	20	-	-	-
FT211L	Principles of food preservation Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT212L	Food Microbiology Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT213L	Food Chemistry Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT214L	Heat Transfer Laboratory	-	-	02	01				IPE	50	20
FT215L	Programming practices for Food Technologists	1	-	02	02				IPE	50	20
	Total	19	1	10	25	-	500	-	-	300	-
HS211	Environmental Studies	02	-	-	-	Project* Theory*	30 70	40	-	-	-
	•	•	•	•	Audit Cours	se-I	•	•	•	•	•
HS212	Soft Skills Development	02	-	-	-	Institute Level	-	-	-	-	-

Total contact hours per week: 30+2+2=34, Total Credits=25

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students.

CIE: Continuous Internal Evaluation SEE: Semester End Examination IPE: Internal Practical Evaluation EPE: External Practical Examination IOE: Internal Oral Evaluation EOE: External Oral Examination

^{*} indicates Environmental Studies project evaluation and the theory examination will be at the end of the year i.e. along with Semester IV End Examination.



DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR SECOND YEAR B.TECH

Scheme of Teaching and Examination: Semester- IV (Food Technology)

Subject	Subject	Tea	(g Sch Credit ırs / V		Examination Scheme (Marks)					
Code						Theory			Practical		
		L	T	P	Total Credits	Scheme	Max. marks	Min. Passing	Scheme	Max. marks	Min. Passing
FT221	Food Process Engineering-I	03	1	-	04	CIE SEE	50 50	20 20	-	-	-
FT222	Food Additives and Contaminants	04	_	-	04	CIE SEE	50 50	20 20	-	-	-
FT223	Food Biochemistry	04	_	_	04	CIE SEE	50 50	20	-	-	-
			_		· ·	CIE	50	20		_	
FT224	Human Nutrition	04		-	04	SEE	50	20	_	_	-
						CIE	50	20	_	-	-
FT225	Fluid Mechanics	04	-	-	04	SEE	50	20	-	-	-
FT221L	Food Process Engineering-I	_	_	02	01	-	-	-	IOE	50	20
	Laboratory			-	V -				EOE	50	20
FT222L	Food Additives and Contaminants Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT223L	Food Biochemistry Laboratory	-	-	02	01	-	-	-	EPE	50	20
FT224L	Human Nutrition Laboratory	-	-	02	01	-	-	-	IOE	50	20
FT225L	Fluid Mechanics Laboratory	-	-	02	01	-	-	-	IOE	50	20
	Total	19	1	10	25	-	500	-	-	300	-
HS221	Environmental Studies	02	-	-	-	Project Theory	30 70	40	-	-	-
			ı		Audit Cours	e-II	•	•	•	•	•
HS222 I	ntroduction to Performing Arts	02	-	-	-	· .	-	-	-	-	-

Total contact hours per week: **30+2+2=34**, Total Credits=25

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students

CIE: Continuous Internal Evaluation SEE: Semester End Examination IPE: Internal Practical Evaluation EPE: External Practical Examination IOE: Internal Oral Evaluation EOE: External Oral Examination

DEPARTMENT OF TECHNOLOGY, SHIVAJI UNIVERSITY KOLHAPUR SECOND YEAR B.TECH

Scheme of Teaching and Examination: Semester- III (Food Technology)

	•	Teac		cheme v urs / W	with Credits (eek)
Subject Code	Subject	L	Т	P	Total Credits
FT211	Principles of Food Preservation	04	-	-	04
FT212	Engineering Mathematics-III	03	01	-	04
FT213	Food Microbiology	04	-	-	04
FT214	Food Chemistry	04	-	-	04
FT215	Heat Transfer	03	_	-	03
FT211L	Principles of food preservation Laboratory	-	-	02	01
FT212L	Food Microbiology Laboratory	-	-	02	01
FT213L	Food Chemistry Laboratory	-	-	02	01
FT214L	Heat Transfer Laboratory	-	-	02	01
FT215L	Programming practices for Food Technologists	1	-	02	02
	Total	20	-	10	25

HS211	Environmental Studies	02	-	-	-	Project* Theory*	30 70	40

	Audit Course				
HS212	Soft Skills Development	02	-	-	Institute Level

Total contact hours per week: 30+2+2=34, Total Credits=25

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students.

^{*} indicates Environmental Studies project evaluation and the theory examination will be at the end of the year i.e. along with Semester IV End Examination.



Scheme of Teaching and Examination: Semester- IV (Food Technology)

		Teaching Scheme with Credits (Hours / Week)					
Subject Code	Subject	L	Т	P	Total Credits		
FT221	Food Process Engineering-I	03	1	-	04		
FT222	Food Additives and Contaminants	04	-	-	04		
FT223	Food Biochemistry	04	-	-	04		
FT224	Human Nutrition	04	-	-	04		
FT225	Fluid Mechanics	04	-	-	04		
FT221L	Food Process Engineering-I Laboratory	-	-	02	01		
FT222L	Food Additives and Contaminants Laboratory	-	-	02	01		
FT223L	Food Biochemistry Laboratory	-	-	02	01		
FT224L	Human Nutrition Laboratory	-	-	02	01		
FT225L	Fluid Mechanics Laboratory	-	-	02	01		
	Total	13	01	10	25		
HS221	Environmental 02 Studies	-		roject neory	30 40 70 40		
	Audit Course				1		
		02	_		1		

	Audit Course				
HS222	Introduction to Performing Arts	02	-	-	Institute Level

Total contact hours per week: **30+2+2=34**, Total Credits=25

Note: Tutorials and Practical to be conducted in batches with batch strength not exceeding 15 students

Detailed Evaluation and Examination Scheme

1. Out of total 100 theory marks, 50 marks are assigned for Continuous Internal Evaluation (CIE). In CIE, obtaining minimum 20 marks is essential. It is similar to term work, the completion of which is mandatory to become eligible to appear for the Semester End Examination (SEE). Failing to complete the term in a particular course i.e. not obtaining 20 marks in CIE out of 50 shall be treated as term not granted in that course and it is on the part of the course teacher to officially inform the particular case through the respective Program Coordinator and the Director to the University Examination Section. The section will take a kind note of the same and it will not issue the hall ticket of the particular students for the SEE in the particular course/s.

2. CIE (50 marks) includes:

- Internal Test I, of 20 marks in 5th week on 1st & 2nd unit
- Internal Test II, of 20 marks in 10th week on 3rd & 4th unit
- Activities for the students: 10 marks. It is at the course owners' discretion to get
 the assignments of varied nature completed by the students. However, the course
 teacher will plan to cover those course objectives that suit course learning
 outcomes and program outcomes that may not be covered in the internal tests.
- 3. For the Semester End Examination (SEE), 100 marks (3 hours) paper will be set and finally it will be converted to 50 marks. The students must secure minimum 40 % i.e. 20 marks in SEE as the University examination passing head.
- 4. Final theory marks (out of 100) will be the addition of CIE (out of 50 marks) and SEE (out of 50 marks).
- 5. Internal Practical/Oral Evaluation (IPE/IOE) will be on the basis of Internal Oral/Practical/Tutorials/Seminar in which students must secure minimum 40% i.e. 20 marks. It is similar to the term work the completion of which is mandatory to be eligible to appear for the Semester End Examination (SEE).
- 6. External Practical/Oral Evaluation (EPE/EOE) will be conducted under the supervision by some external course expert. The minimum score 40% i.e. 20 marks is required to be secured as the University's passing head in EPE/EOE.

- 7. *Semester End Examination duration will be 4 hrs.
- 8. Equivalence for the Course: As elaborated at the end of this whole curriculum document.

Academic Autonomy:

- 1. Flexibility in deciding Structure and Contents of Curriculum with reasonable frequency for changes in the same.
- 2. Continuous Assessment of Students performance with newly adopted Credit System based on award of grade.
- 3. Credits are simply a means of attaching relative values to courses of different components. These are a currency of learning and in general regarded as a measure of the time typically required to achieve a given curricular outcome.
- 4. All courses (Courses) under each Program/Discipline are unitized.

Credit system:

Education at the Institute is organized around the semester-based credit system of study.

The prominent features of the credit system are a process of continuous evaluation of a student's

performance/progress and flexibility to allow him/her to progress at an optimum pace suited to

his/her ability or convenience. Each course by every student needs to fulfill minimum

requirements of credits for continuation.

A student's performance/progress is measured by the number of credits that he/she has

earned, i.e. completed satisfactorily. Based on the course credits and grades obtained by the

student, grade point average is calculated. A minimum grade point average is required to be

maintained for satisfactory progress and continuation in the Program. Also a minimum number

of earned credits and a minimum grade point average should be acquired in order to qualify for

the degree. All Programs are defined by the total credit requirement and a pattern of credit

distribution over courses of different categories.

Course credits assignment:

Each course, except a few special courses, has a certain number of credits assigned to it

depending upon its lecture, tutorial and laboratory contact hours in a week. This weightage is

also indicative of the academic expectation that includes in-class contact and self-study outside

of class hours.

Lectures and Tutorials: One lecture or tutorial hour per week per semester is assigned one

credit.

Practical/Laboratory: One laboratory hour per week per semester is assigned half credit.

Example: Course: Chemistry-I: 5 credits (4-0-2)

The credits indicated for this course are computed as follows:

4 hours/week lectures = 4 credits

0 hours/week tutorial = 0 credit

2 hours/week practical = $2 \times 0.5 = 1$ credit

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The contact hours in this case of **5** credits course is 6 hours per week. (**4** h Lectures + **0** h Tutorial + **2** h Practical=6 hours per week.)

For each lecture or tutorial credit, the self study component is 1 hour/week and 2 hours/week. In the above example, the student is expected to devote 3 + 1 = 4 hours per week on self study for this course, in addition to class contact of 5 hours per week.

Earning credits:

At the end of every course, a letter grade is awarded in each course for which a student had registered. On obtaining a pass grade, the student accumulates the course credits as earned credits. A student's performance is measured by the number of credits that he/she has earned and by the weighted grade point average.

The credit system enables continuous evaluation of a student's performance and allows the students to progress at an optimum pace suited to individual ability and convenience.

Features of Credit System at Shivaji University, Kolhapur:

Every course is allotted credits based on its academic importance/weightage.

- 1. All Courses may not have same credits.
- 2. There will be 23 to 28 Credits / Semester.
- 3. Absolute Grading System with 7 Passing Grades viz. AA, AB, BB, BC, CC, CD, DD and FF for failure.
- 4. Getting FF grade in 4 heads in one academic year, he/she is considered as failed.
- 5. Continuous Evaluation: Unit Test I i.e. T₁ [20 marks], and Unit Test II i.e. T₂ [20 marks]. Activities will be for 10 marks and the course owner/in charge will have discretion to decide the nature of activities.
- 6. Standardization of courses: Each course is unitized in 6 numbers. Unit Test I on units I and II while Unit Test II on units III & IV, SEE will be based on all the units of the course curriculum.
- 7. Unit Test I & Unit Test II will be supervised and evaluated by internal course teachers while SEE will be evaluated mostly by external and internal teachers as joint examiner ships.

- 8. Any request for re-test will not be entertained after internal test.
- 9. For both the semesters' failure courses, re-examination will be only after the even Semester End Examination. No re-examination will be conducted for odd semester courses in even semester or vice-versa.

Attendance rule:

All students must attend every lecture, tutorial and practical class. However, to account

for late registration, sickness or other such conditions, the attendance requirement will be a

minimum of 75 % of the classes actually held. A student with less than 75 % attendance in a

course during the semester, in lectures, tutorials and practical taken together (as applicable), will

be awarded the 'F' grade in that course irrespective of his/her performance in the tests.

Taking into account the consolidated attendance record for the whole semester, the

course in charge in consultation with the Program Coordinator will award 'XX' grade to the

student who is deficient in attendance. For the purpose of attendance calculation, every

scheduled practical class will be counted as one unit irrespective of the number of contact hours.

Attendance record will be maintained based upon roll calls (or any equivalent operation)

in every scheduled lecture, tutorial and practical class. The course owner will maintain and

consolidate attendance record for the course (lectures, tutorials and practical together, as

applicable).

Evaluation system:

1. Semester Grade Point Average (SGPA) =

 \sum (course credits in passed courses X earned grade points)

 \sum (Course credits in registered courses)

2. Cumulative Grade Point Average (CGPA) =

 \sum (course credits in passed courses X earned grade points) of all Semesters

 \sum (Course credits in registered courses) of all Semesters

3. At the end of B. Tech Program, student will be placed in any one of the divisions as

detailed below:

Ist Division with distinction: CGPA \geq 8.25 and above

Ist Division

: CGPA > 6.75 and < 8.25

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IInd Division : CGPA
$$\geq$$
 6.25 and $<$ 6.75

As per AICTE Handbook (2011-12), gradation is as follows:

Grade Points	Equivalent Percentage Range
6.25	55
6.75	60
7.25	65
7.75	70
8.25	75

Conversion of CGPA to corresponding equivalent percentage marks for CGPA>5.0 may be obtained using the following equation:

Equivalent Percentage marks = (Respective CGPA x 10) – 7.5

An example of these calculations is given below:

Typical academic performance calculations - I semester

Course no.	Course credits	Grade awarded	Earned credits	Grade points	Points Secured
Col 1	Col 2	Col 3	Col 4	Col 5	Col 6
					(Col 4* Col 5)
MALXXX	5	CC	5	6	30
CSLXXX	4	CD	4	5	20
PHLXXX	4	AA	4	10	40
PHPXXX	2	BB	2	8	16
MELXXX	4	FF	0	0	0
TTNXXX	2	AB	2	9	18
Total	21		17	38	124

1. Semester Grade Point Average (SGPA) =

2. Cumulative Grade Point Average (CGPA) =

Cumulative points earned in all passed courses = 124 (past semesters) + 124 (this sem.) = 248 Cumulative earned credits = 23 (past semesters) + 21 (this sem.) = 44

$$\sum (124 + 124)$$

$$\sum (23 + 21)$$
= 5.63

Chart for marks range and its corresponding grade and grade points

Marks Range	Grade Points	Grade	Description of Performance
91-100	10	AA	Outstanding
86-90	09	AB	Excellent
76-85	08	BB	Very Good
66-75	07	ВС	Good
56-65	06	CC	Fair
46-55	05	CD	Average
40-45	04	DD	Poor
Below 40	00	FF	Fail
		\$	Passed in first attempt
		PP	Passed (Audit Course)
		NP	Not Passed (Audit Course)
		** 2 nd *** 3 rd *** 4 th	One grade punishment for 2 nd ,3 rd , 4 th ,attempt,

Audit Courses:

Additional courses shall be included as audit courses from the third semester onwards. While the performance of the student in audited courses shall be included in the Grade Card, these grades do not contribute to SGPA or CGPA of the concerned student.

Award of Degree:

Following rules prevail for the award of degree:

- 1. A Student has registered and passed all the prescribed courses under the general institutional and departmental requirements.
- 2. A student has obtained CGPA \geq 4.5.

- 3. A student has paid all the institute dues and satisfied all the requirements prescribed.
- 4. A student has no case of indiscipline pending against him/her.
- 5. Institute authorities shall recommend the award of B.Tech degree to a student who is declared to be eligible and qualified for above norms.

CGPA Improvement Policy for award of degree:

An opportunity shall be given to a student who has earned all the credits required by the respective program with CGPA greater than or equal to 4.00 but less than 4.50, to improve his/her grade by allowing him/her to appear for 100% examinations of maximum two theory courses of seventh and eighth semester. However, CGPA shall be limited to 4.5 even though the performance of a student as calculated through modified CGPA becomes greater than 4.5.

B. Tech (Food Technology) Program Educational Objectives (PEOs), Program Outcomes (POs) and Program Specific Outcomes (PSOs) of the Program:

	Program Specific Outcomes (PSOs) of the Program:
	Program Educational Objectives (PEOs):
	Students are expected to have strong theoretical and practical knowledge in the field of
PEO1	Food Engineering and Technology to cater the need of society by Integrating
	engineering and basic sciences
	To produce competent graduates who shall Pursue careers in the field of food
PEO2	processing, quality control, new product development and techno-marketing
	Graduates are expected to be equipped with innovative ideas, project management
PEO3	skills which will make them capable to start-up and grow as an entrepreneur
	To prepare graduates who will apply the technical knowledge and know-how to solve
PEO4	the problems related to food processing and preservation for the benefit of local
	region and of the country at large
PEO5	To get ready graduates who will be effective experts in government, academia,
1EO3	research and consulting firms in the field of food engineering and technology
	Program Outcomes (POs)
	Acquire in depth knowledge, understanding and skills through competency in core
PO1	areas: includes food chemistry, food microbiology, food engineering, food processing,
	nutrition and food safety and quality.
	Graduates will have an ability to know the fundamental chemistry with respect to
PO2	properties and reactions of various food components and will have sufficient
	knowledge to control reactions in foods during processing and storage
	Graduates will have an ability to spot and resolve the problems in the actual situation
PO3	during food processing, food quality controlling, food packaging and storage
PO4	Graduates will have an ability to express practical proficiency in a food analysis
104	laboratory
PO5	Graduates will have a knowledge of food pathogens and spoilage microorganisms and

	will have an ability to inactivate or killing of food microorganism or food pathogens
PO6	Graduates will have an ability of designing and development of food products as per need of society
PO7	Graduates will have an ability to communicate effectively for self development
PO8	Graduates will have know ledge of quality management system and will have an ability to apply in food processing industries
PO9	Graduates will have knowledge of rules and regulations of food hygiene and sanitation and will able to apply for safe and nutritious food processing
PO10	Graduate will have knowledge and ability of industrial economics and management of food industries
PO11	Graduates will be recognized as food technologists as a responsible technocrats who impact society through their knowledge and actions
PO12	Graduates will have a perceptive of professional and ethical responsibility
	Program Specific Outcomes(PSOs)
PSO1	Graduates will apply the knowledge of food chemistry, food preservation, food processing and food packaging for the effective utilization of agricultural commodities to develop healthy and nutritious food
PSO2	Graduates will design the equipments for the modernization of traditional food process methods
PSO3	Graduates will apply principles from the various aspects of food technology and related disciplines to solve practical and real-world problems
PSO4	Gradates will able to develop quality food processes by considering cost, safety and health aspects

Class & Semester	:	S. Y. B. Tech (Food Technology)	Part II, Semester III			
Course Title	:	Principles of Food Preservation	Course Code:	:	FT211	
Teaching Scheme (Hours)	:	Lectures 4 hours/weeks=4 x 13 weeks= 52 hours minimum Tutorial= 0 hour/week Practical= 02 hours/week	Total Credits	:	04+01=05	
Evaluation Scheme (Marks)	:	CIE = 50	Duration of SEE		3 hours	
Revision:	:	Third	Month	:	December 2016	

Pre-requisites	:	In order to complete the course studies successfully, it is important to have a good command of English and knowledge of basic science.
Type of Course	:	Theory
Course Domain	:	Core
	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate
Skills Imbibed		Affective: Awareness, Respond, Value, Organize
		Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

- 1. This course helps the students to apply basic food science knowledge and get to know biochemical changes occurring during various processing and preservation techniques.
- 2. Introduce students to methods of preserving food to avoid losses.
- 3. Inform students on the technical mechanism of food preservation.
- 4. Introduce students to various methods of food processing and preservation.

Course Outcomes:

- 1. Understand the need for food processing.
- 2. Understand the various food processing techniques.
- 3. Understand the different preservation technique.
- 4. Identify the major and minor constituents of food and the chemical reactions in which they participate.
- 5. Describe the principles involved in the processing of the major types of food products.
- 6. Understand the principles of food spoilage and the ways to prevent.

	Hours					
Unit I: Introduction Scope and Importance of food processing, National and International perspectives, Objectives and techniques of food preservation.						
UNIT II Food preservation by low temperature						
Cooling, Refrigeration, freezing and freeze drying	8					
UNIT III Food preservation by heating Blanching, pasteurization, sterilization, UHT processing, extrusion cooking of food.	8					
UNIT IV Preservation by drying Processing and preservation by drying, concentration and evaporation-types of dryers and their suitability for different food products;	9					
UNIT V Food preservation by Nonthermal method Chemical preservation, fermentation methods for food preservation, irradiation, membrane technology.	9					
UNIT VI Recent methods for food preservation Pulsed electric field processing, high pressure processing, processing by using ultrasound, dielectric, ohmic and infrared heating etc.	10					
Text Books :						
1. Manoranjan Kalia and Sangita Sood, "Food preservation and processing", Kalyani Publishers. New 2004. 2.	Delhi					
2. Richardson, T. and Finley, J.W, "Chemical changes in food during processing". Macmillon Publish Canada 2003.	hers.					
Reference Books :						
1. Fellows, P. and Ellis H. 1990"Food processing technology: principles and practice", Wood Head						
Publishing Ltd.						
2. Jelen, P. 1985. Introduction to Food Processing. Prentice Hall, Reston Virginia, USA.						
3. Heldman, D.R. and Singh R. P Ed.IV 2009. Introduction to Food Engineering. Elsevier Pub						
4. Desrosier, N.W, "The Technology of Food Preservation", CBS Publishers and Distributors, New Des	elhi					
1996.						

Class & Semester	:	S. Y. B	. Tech	(Food	Te	chnology),	Part II, Semest	er	III
		Principle	of Foo	d Prese	rva	tion			
Course Title	:	Laborate		u I I ese	ı va	uon	Course Code:	:	FT211L
Teaching Scheme (Hours)	:	2 hr /weel	k= 2 x13	ırs		Credits	:	1	
Evaluation Scheme (Marks)	:	IPE :		EPE EOE	:	50 Nil	Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third					Month	:	December 2016

Pre-requisites

• In order to complete the course studies successfully, it is important to have a good command of English and knowledge of basic science.

Type of Course

: Practical

Course Domain

: Core

Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed

Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 1. To impart knowledge and skills on processing and preservation methods for wholesome food products.
- 2. Preserve food by drying, freezing, water-bath canning and other methods.
- 3. Calculate the value of the food preserved.
- 4. Evaluate preserved products.

Course Outcomes:

- 1. To identify & select processing equipment and preservation methods appropriate for specific foods.
- 2. To describe the effects of preservation methods on the quality of food.
- 3. Food Product Development Exercise
- 4. To implement preservation methods that make use of heat/cold, drying, acid, added chemicals, controlled air, pressure, and high energy radiation.
- 5. To identify indirect approaches to food preservation: packaging, hygienic design, sanitation, GMP
- 6. To use SOPs and SSOPs during laboratory exercise.
- 7. To identify quality-loss mechanisms as biological, chemical, and physical.

Practical List

- 1. Demonstration of various machineries used in processing
- 2. Demonstration of effect of blanching on quality of foods
- 3. Preservation of food by heat treatment- canning of fruits and vegetables
- 4. Preservation of food by sugar preparation of jam
- 5. Preservation of food by salt pickle
- 6. Preservation of food by chemicals squash/ketchup/cordial
- 7. Drying of pineapple slices, apple slices in cabinet drier
- 8. Drying of green leafy vegetables
- 9. Drying of foods using freeze-drying process
- 10. Food preservation by maintaing anaerobic condition
- 11. Osmotic dehydration of fruit, canning of fruits and vegetables
- 12. Demonstration of preserving foods under cold v/s freezing process
- 13. Visit to canning or food processing Industry

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. Fellows, P. and Ellis H. 1990"Food processing technology: principles and practice", Wood Head Publishing Ltd.
- 2. Jelen, P. 1985. Introduction to Food Processing. Prentice Hall, Reston Virginia, USA.
- 3. Heldman, D.R. and Singh R. P.. Ed.IV 2009. Introduction to Food Engineering. Elsevier Pub
- 4. Desrosier, N.W, "The Technology of Food Preservation", CBS Publishers and Distributors, New Delhi 1996.

Class & Semester	:	S. Y. B.Tech (F	Food T	Part II, Sem III				
Course Title	:	Engineering Mathe Lectures= 3 hours/v		III		Course Code:	:	FT212
Teaching Scheme (Hours)	:	=3 x 13 weeks= 39 h Tutorial= 01 hrs/v =3 x 13 weeks= 13 h Practical= 00 hrs/v	week hrs minin			Total Credits	:	03+ 01+ 00 =04
Evaluation Scheme (Marks)	:	CIE = 50 IO SEE = 50	E=00 DE=50 PE=00	:	Grand Total= 100	Duration of SEE	:	3 hrs
Revision:	:	Third				Month	:	December 2016

Pre-requisites	:	Engineering Mathematics-I and Engineering Mathematics-II
Type of Course	••	Theory
Course Domain	:	Core
Skills Imbibed	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate Affective: Awareness, Respond, Value, Organize Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination (SEE)

Course Objectives:

- 1. To describe solution of LDE and its applications.
- 2. To introduce Laplace Transform.

- 3. To introduce Partial Differential Equations and its Applications.
- 4. To introduce Probability and its Applications.
- 5. To explain Vector Differentiation and Vector Integration
- 6. The student must be able to formulate a mathematical model of a real life and engineering problem, solve and interpret the solution in real world.

Course Outcomes:

At the end of course student will able to

- 1. Solve Linear Differential Equations and apply them to realistic problems.
- 2. Solve Partial Differential Equations for solving problems in Food Engineering.
- 3. Understand Application of Laplace transform and Probability in Process Instrumentation Dynamics and Control.
- 4. Apply knowledge of Vector Calculus to solve engineering problems.

Curriculum Content	Hours
Unit: I Linear Differential Equations:	07
Linear Differential Equations with constant coefficients, Homogenous Linear	ı
differential equations, method of variation of parameters	İ
Unit: II Laplace Transform:	07
Definition, properties of Laplace transforms, transforms of derivatives, transforms of integral, Inverse Laplace transforms, Convolution theorem. Applications to initial value boundary problems, Heaviside Unit step function, Diracdelta function, Periodic function.	
Unit: III Partial differential equations:	06
Four standard forms of partial differential equations of first order.	
Unit: IV Application of Partial differential Equations Classification of PDE, Solution of Wave Equation, One dimensional heat equation and two dimensional Laplace equation by the method of separation of variables, use of Fourier series.	06
Unit :V Probability:	06
Random variable, Probability mass function and probability density function, Binomial, Poisson and Normal distributions.	
Unit : VI Vector Calculus:	l
Vector Differentiation: Differentiation of vectors, Gradient of scalar point function,	07
Directional derivative, Divergence of vector point function, Curl of a vector point	İ
function. Irrotational and solenoidal vector field.	1
Vector Integration: The line integral, Surface integral, volume integral, Gauss's	1

Divergence theorem, Stoke's theorem, Green's theorem (Without proof).

Suggested list of Tutorials/Assignments-

- 1. To find solution of LDE with constant coefficients
- 2. Applications of LDE
- 3. Laplace Transform
- 4. Partial Differential Equations
- 5. Applications Of PDE
- 6. Probability
- 7. Vector differentiation
- 8. Vector Integration

General Instructions:

- 1. Batch wise tutorials are to be conducted. The number of students per batch should be as per the practical batches.
- **2.** Students must be encouraged to solve engineering mathematics problems using different software's in tutorial class only.
- 3. Each Student has to write at least 6 assignments on entire syllabus.

Text Books

:

1. Erwin Kreyszig, "Advanced Engineering Mathematics (7th Edition)", Wiley Eastern Ltd., Bombay.

Reference Books

:

- 1. B. S. Grewal, "Higher Engineering Mathematics", Khanna Publications, New Delhi.
- 2. C.R.Wylie, "Advanced Engineering Mathematics", McGraw Hill Publication, New Delhi.
- 3. Merle C. Potter, "Advanced Engineering Mathematics", OXFORD University Press, 3rd Edition
- 4. P. N. Wartikar and J. N. Wartikar, "A Text Book of Engineering Mathematics (Volume-I ,II & II)", Pune Vidyarthi Griha Prakashan, Pune.
- 5. Shanti Narayan, "Differential Calculus" S. Chand and company, New Delhi.
- 6. S. S. Sastry, "Engineering Mathematics (Volume-I)", Prentice Hall Publication, New Delhi.
- 7. B.V. Ramana, "Higher Engineering Mathematics", Tata McGraw-Hill.
- 8. M. D. Greenberg, "Advanced Engineering Mathematics", Pearson Education.
- 9. H. K. Das, "Advanced Engineering Mathematics", S. Chand Publication.

Class & Semester	:	S. Y. B.	Tech (Food	ΙT	echnology)	Part II, Semester III		
Course Title	:	Food Mic	robiology			Course Code:		FT 213
Teaching Scheme (Hours)	:	minimum Tutorial=	eeks=4 x 13 v 0 hour/week 02 hours/weel		ks= 52 hours	Total Credits	:	04+01 =05
Evaluation Scheme (Marks)	:	CIE = 50 SEE = 50	IPE=Nil IOE=Nil EPE= 50	:	Grand Total=150	Duration of SEE	:	3 hours
Revision:	:	Third	•			Month	:	December 2016

Pre-requisites	:	In order to complete the course studies successfully, it is important to have a good command of English and basic knowledge of general microbiology
Type of Course	:	Theory
Course Domain	:	Core
Skills Imbibed	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate Affective: Awareness, Respond, Value, Organize Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

On completion of the course the student are expected to

- 1. Be able to understand and identify the various microbes associated with foods and food groups.
- 2. Be able to understand and identify the role of this microbe in food spoilage, food preservation.
- 3. Understand the role of pathogens in food borne infections.
- 4. Understand the method used to detect pathogens in foods.

Course Outcomes:

- 1. Better understanding of various microbes associated with food and food groups.
- 2. Familiarize the role of microbes in food spoilage and food preservation.
- 3. Obtain Knowledge in methods to detect pathogens in food.
- 4. Describe the beneficial role of microorganisms in fermented foods and in food processing.
- 5. Utilize laboratory techniques to detect, quantify, and identify microorganisms in foods.
- 6. Acquire, discover, and apply the theories and principles of food microbiology in practical, real-world situations and problems.
- 7. Develop success skills in communication, critical thinking, interaction, information acquisition and interpretation, organization, professionalism, leadership, auto-didactics and life-long-learning.

UNIT I Introduction to microbiology	Hours
Evaluation & Scope of microbiology. General Microbiology, Cultural characteristics and reproduction of bacteria, yeasts, fungi, actinomycetes, algae, protozoa and rickettsia. Genetic recombination, transformation and bacterial conjugation, mutation & mutagenesis.	10
UNIT II Growth curve	
Physical & Chemical factors affecting growth and destruction of micro-organisms, growth curve.	8
UNIT III Viruses	
Structure & reproduction with particular reference to food born viruses.	8
UNIT IV Food Preservation	
Principles of Food Preservation, Methods of preservations, Drying, Dehydration, Freezing, Chemical Preservation, Mechanical destruction & Maintenance of anaerobic condition.	10
UNIT V Spoilage of Food	
Microbial Spoilage & cereals & cereal products, milk & milk products, fruit & Vegetable products, meat, poultry egg & fish products, sugar & Sugar Products.	10

UNIT VI Food noisoning									
UNIT VI Food poisoning									
Intoxication, Food borne illness									
Text Books	:								
1. Prescott Dunn,	" Ir	dustrial Microbiology" CBS Publisher							
2. Purohit S. S. "N	/lici	robiology fundamentals and applications" Edition, 6. Publisher, Agrobios, 2003							
Reference Books	:								
1. Frazier, W.C., a	ind	Westhoff, D.C. 1988. Food Microbiology, 4th ed. McGraw-Hill, New York.							
2. Jay, J. M. 2000. Modern Food Microbiology. 6th ed. Chapman & Hall. New York, N.Y.									
	3. Mossel, D.A.A., Corry, J. E. L., Struijk, C. B., and Baird, R. M. 1995. Essentials of the Microbiology of Foods. John Wiley & Sons. New York, NY.								

Class & Semester	:	S. Y. B	S. Y. B. Tech (Food Technology), Part II, Semester III								
Course Title	:	Food Mi	crobiol	logy Lal	bor	atory	Course Code:	:	FT 212L		
Teaching Scheme (Hours)	:	2 hr /weel	k= 2 x13	3= 26 ho	urs		Credits	:	1		
Evaluation Scheme (Marks)	:	IPE :	Nil Nil	EPE EOE	:	50 Nil	Duration of Exam (in case of External Evaluation)	:	02 hours		
Revision:	:	Third					Month	:	December 2016		

Pre-requisites

• Must be capable of safe lab work with cultures.

Type of Course

Practical

Course Domain

Core

•

Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed

Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 1. Enable students to understand the methods of isolating and characterizing various microbes associated with foods and food groups.
- 2. Enable students to understand and use various microbiological techniques for the study of foods.
- 3. Understand the methods used to detect pathogens in foods.

- 1. Complete understanding of isolation, characterization of various microbes associated with foods and food groups.
- 2. Familiarize with microbiological techniques for the study of foods.
- 3. Better understanding of methods to detect pathogens in foods.
- 4. Students should be able to significance and activities of microorganisms in food.
- 5. Students should be able to microbiology of different types of food commodities.
- 6. Students should be able to: Describe the characteristics of food borne, waterborne and spoilage microorganisms, and methods for their isolation, detection and identification.

Practical List

- 1. Study of instruments used for microbiology, cleaning and sterilization of glassware.
- 2. Preparation of media, techniques of incubation,
- 3. Staining methods (monochrome staining, gram staining, flagella staining, capsule staining and endo spore staining),
- 4. Quantification of Microbes: Sampling and Serial Dilution; Bacterial count in food products—TVC
- 5. Pure culture techniques (streak plate/pour plate), growth curve.
- 6. Isolation of molds from foods, microbial examination of cereal and cereal products.
- 7. Microbial examination of fruits and vegetables.
- 8. Microbial examination of milk and milk products, microbial examination of sugar, salt and spices.
- 9. Microbial examination of meat and meat products.
- 10. Microbial examination of water.

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- 11. Microbial examination of fermented food.
- 12. Enumeration & Isolation of Staphylococci from ready to eat street foods
- 13. Effect of cleaning and disinfection on microbial load Effect of cleaning and disinfection on microbial load.

Lab Manual

1. I Harrigan, W.F. "Laboratory Methods in Food Microbiology" Academic Press, 2011.

Reference Books

- 1. Frazier, W.C., and Westhoff, D.C. 1988. Food Microbiology, 4th ed. McGraw-Hill, New York.
- 2. Jay, J. M. 2000. Modern Food Microbiology. 6th ed. Chapman & Hall. New York, N.Y.
- 3. Mossel, D.A.A., Corry, J. E. L., Struijk, C. B., and Baird, R. M. 1995. Essentials of the Microbiology of Foods. John Wiley & Sons. New York, NY.
- 4. Prescott Dunn, "Industrial Microbiology" CBS Publisher
- 5. Purohit S. S. "Microbiology fundamentals and applications" Edition, 6. Publisher, Agrobios, 2003

Class & : S. Y. B.Tech (Food Technology) Part II, Semester III

Semester

Course Title : FOOD CHEMISTRY Code: : FT 214

Lectures

Teaching 4 hours/weeks=4 x 13 weeks= 52 hours

Scheme : minimum : 04+01 =05

(Hours) Tutorial= 0 hour/week

Practical= 02 hours/week

Evaluation CIE = 50 IPE=Nil : Grand Duration of

Scheme : (Marks) : SEE = 50 | IOE=Nil : Total=150 | SEE

Revision: : Third Month : December

2016

: In order to complete the course studies successfully, it is

Pre-requisites important to have a good command on English and Basic

knowledge of chemistry

Type of Course : Theory

Course Domain : Core

: Cognitive: Recall, Understand, Apply, Analyze,

Synthesize, Evaluate

Skills Imbibed Affective: Awareness, Respond, Value, Organize

Psychomotor: Imitation, manipulation, articulation,

naturalization

Course Assessment Methods:

1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments

2. Semester End Examination.

Course Objectives:

To enable the students to understand

- 1. The chemistry underlying the properties and reactions of various food components and their role in human nutrition
- 2. The chemistry of food to control a chemical and biochemical reaction that

3 hours

- influence food quality
- 3. The principles behind analytical techniques associated with food components and related problems
- 4. The role of food nutrients and its use for preservation of food (concepts emphasized in a laboratory experiment)
- 5. The function of food enzymes; effect of pH, temperature, water activity, ionic strength; applications of chymosin and amylases as food processing aids
- 6. The functional role of food components and their interaction in food products in terms of colour, flavour, texture and nutrient composition

Course Outcomes:

Upon successful completion of this course, students will be able to:

- 1. Recognize, distinguish and describe the molecular structures and properties of major food components.
- 2. Describe major food chemical reactions and their mechanisms.
- 3. Relate key chemical groups on food molecules to their role in common reaction mechanisms of importance in foods.
- 4. Analyze and predict how the composition of foods with regard to carbohydrates, lipids, protein and water influence their stability.
- 5. Examine and assess implications for food formulations for achieving objectives of food quality and palatability, cost and healthfulness.
- 6. Analyze and interpret the role of food chemistry in practical food situations.
- 7. Assess and evaluate the implications of food chemistry principles for current topics of importance in the food industry and for consumers.

Hours **UNIT I Introduction to Food Chemistry** 5 Importance, Scope and Recent Development **UNIT II Water** Role and types of water in foods, functional properties of water activity and sorption 7 isotherm. **UNIT III Carbohydrate** Chemistry of carbohydrates (Classification, Structure and Sources and properties), 9 Functional Characteristics of carbohydrates and chemical changes, modification of carbohydrate. **UNIT IV Protein** Chemistry of proteins (Classification, Structure and Sources and properties) Functional

characteristics of protein and chemical changes in protein, modification of protein,

enzymatic and non-enzymatic browning

9

UNIT V Lipids 10

Chemistry of lipids (Classification, Structure and Sources and properties) Functional characteristics of

lipids and chemical changes in lipids, hydrogenation of fat.

12

UNIT VI Vitamins & Minerals

Vitamins: Classification of Vitamins, Structure of Vitamins, Sources of Vitamins, properties, effect of processing on Vitamins and deficiency problems of Vitamins.

Minerals: Classification, Structure, Sources and properties of minerals, effect of processing on of Minerals and deficiency problems of minerals.

Text Books

- 1. Belitz H.-D, Grosch W and Schieberle P. Food Chemistry, 3rd Revised Edition, Springer-Verlag, 2004.
- 2. Meyer, Lillian Hoagland "Food Chemistry". CBS Publishers, 1987.
- 3. deMan, John M. "Principles of Food Chemistry". 3rd Edition, Springer, 1999.
- 4. Chopra, H.K. and P.S. Panesar. "Food Chemistry". Narosa, 2010.

Reference Books

- 1. Belitz, H.D.. Grosch "Food Chemistry" 3rd revised ed. Springer Berlin, Heidelberg, New York.
- 2. O.R.Fennema "Food Chemistry" Marcel Dekker, Inc., New York
- 3. Mayer Liiiian H. "Food Chemistry" AVI Publishing Co., Westport, CT, 1978, 3rd ed
- 4. Introductory foods, Bennion M. and Hughes, D. (1975), Macmillan publishing Co., New York.
- 5. Food facts and principles, Sakuntala Manay and shadaksaraswamy, M (1987) Allied Publishers, New Delhi.

Class & Semester	:	S. Y.]	S. Y. B. Tech (Food Technology), Part II, Semester III							
Course Title	:			HEMI ATOR	STRY Y	Course Code:	:	FT 213L		
Teaching Scheme (Hours)	:	2 hr /v	vee	k= 2 x	13= 26 h	ou	Credits	:	1	
Evaluation Scheme (Marks)	:	IPE IOE						Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third	•	•	•	•		Month	:	December 2016

Pre-requisites

: In order to complete the course studies successfully, it is important to have a good command on English and subject proficiency.

Type of Course

: Practical

Course Domain

: Core

Skills Imbibed

: Cognitive: Understand, Apply, Analyze, Evaluate, Create Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 1. To impart practical knowledge and skills.
- 2. To study and understand the physical and chemical properties of foods
- 3. To study and calculate the value of food components
- 4. To study the basic nutrients and their requirements for human nutrition
- 5. Evaluate new product development.
- 6. Demonstrate practical proficiency in a food analysis laboratory.

- 1. Better understanding of physical and chemical properties of food.
- 2. Familiarize in precipitation of casein and gellation of starch.
- 3. Laboratory exercises on specific analytical problems.

- 4. Understand proper use of methods of analytical analyses
- 5. Understand various methodologies for analysis of various components in foods
- 6. Understand how government regulations affect analysis of components in food.

Practical List

- 1. Determination of moisture content.
- 2. Qualitative test for carbohydrates, Qualitative tests for amino acid, Qualitative tests for lipid.
- 3. Estimation of crude fat.
- 4. Estimation of crude protein by microkildhal method.
- 5. Determination of reducing sugar by filling method.
- 6. Determination of total sugar by filling method.
- 7. Determination of acid value, saponification volume and iodine number of fats.
- 8. Determination of Ca, Iron, phosphorus and Ash content of food.
- 9. Estimation of Viscosity of foods.
- 10. Properties of solutions- sugar & salt
- 11. Preparation of emulsions.
- 12. Foaming properties of proteins.
- 13. Iso-electric precipitation of casein, Effect of rennin on milk proteins.

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. Belitz, H.D.. Grosch "Food Chemistry" 3rd revised ed. Springer Berlin, Heidelberg, New York.
- 2. O.R.Fennema "Food Chemistry" Marcel Dekker, Inc., New York
- 3. Mayer Liiiian H. "Food Chemistry" AVI Publishing Co., Westport, CT, 1978, 3rd ed
- 4. Introductory foods, Bennion M. and Hughes, D. (1975), Macmillan publishing Co., New York.
- 5. Food facts and principles, Sakuntala Manay and shadaksaraswamy, M (1987) Allied Publishers, New Delhi.

Class & Semester	:	S. Y. B. T	ech (Food T	ec	hnology) Pa	rt II, Seme	ster	· III
Course Title	:	Heat Transfe	er		Course Code:	:	FT215	
Teaching Scheme (Hours)	:	minimum Tutorial= 00	eks=3 x 13 v 0 hour/week 2 hours/week	Total Credits	:	03+01 =04		
Evaluation Scheme (Marks)	:	CIE = 50 SEE = 50	IPE=50 IOE=Nil EPE= Nil	:	Grand Total=150	Duration of SEE	:	3 hours
Revision:	:	Third		•		Month	:	December 2016

Pre-requisites

In order to complete the course studies successfully, it is important to have a good command of English. Other Pre-requisites include Engineering Physics, Chemistry-I, and Fluid Flow Operations.

Type of Course

Theory

Course Domain

Core

Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate

Skills Imbibed

Affective: Awareness, Respond, Value, Organize Psychomotor: Imitation,

manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

- 1. To introduce the course with respect to importance of professional ethics, engineering design, communications and teamwork, use of modern tools and life-long learning and further to understand, identify and explain three modes of heat transfer namely conduction, convection and radiation.
- 2. To formulate the equations for calculating heat flux for conduction, convection, radiation, boiling, condensation.
- 3. To apply knowledge and information gained on design and analysis of Heat transfer systems, further to interpret and select the type of heat transfer system and to evaluate heat transfer rates.
- 4. To understand phase-change phenomena and latent heat of vaporization, including free convective, nucleate and film boiling, as well as drop wise and film condensation.
- 5. To understand, distinguish and describe different types of evaporators further to assess, select evaporators for specific duty and to improve their performance through heat transfer knowledge

6. To construct the governing equations for designing and analyzing heat transfer equipment and to further make selection of heat exchangers also to think of possible rearranging or redesign of the heat transfer systems to save energy.

Course Outcomes:

- 1. Learn fundamentals of heat transfer
- 2. Identify principles of different modes of heat transfer
- 3. Design and analyze heat transfer operations and equipment
- 4. Compare performances and select type of heat transfer equipment
- 5. Learn industrial applications of heat exchangers and evaporators
- 6. Solve heat transfer problems of different difficulty levels through tutorials
- 7. Complete design of a heat exchanger through assignments/ group task
- 8. Understand importance of heat transfer with respect to societal aspects, environmental concerns, professional ethics, team work, interdisciplinary tasks

Curriculum Content Hours **Unit I: Introduction Heat Transfer** Introduction to three modes of heat transfer: Conduction, convection & radiation. General laws of 4

Unit II: Conduction:

heat transfer.

Fourier's law, Thermal Conductivity- its variation with temperature and Pressure and its relationship with electrical conductivity. Heat transfer through composite walls and cylinders. Unsteady state heat transfer through some important shapes. Different types of insulating materials, general properties & application of insulators.

Unit III: Natural and Forced Convection Natural convection from vertical plates and horizontal cylinders. Forced convection: In laminar flow-Heat transfer in plate and tubes. In turbulent flow-Empirical equations for individual coefficients: inside tubes, outside tubes, outside bundle of tubes, flow past spheres. Significance of Prandtl number, Nusselt number, Grashof number, Graetz number and Peclet

number etc. Correction for tube length. Corrections for heating and cooling the fluid. Various analogies between heat & momentum transfer.

Unit IV: Radiation Radiation laws like Stefan Boltzmann's law, Kirchhoff's law, Wien's law, Plank's law etc. Black body, Grey body. Transmissivity, Absorptivity, Reflectivity, Emissivity of black bodies and gray bodies. Application of thermal radiation: Radiation Transfer between surfaces. Radiation through

semi transparent materials. Unit V: Heat transfer with phase change

Boiling of liquids, Pool boiling curve, different types of pool boiling, Condensation of vapor, film wise and drop wise condensation, weighted LMTD & Overall Heat transfer Coefficient for de superheating & sub cooling.

Unit VI: Heat Exchange equipment

Double pipe heat exchangers. Individual and overall heat transfer coefficient, LMTD, Variable

7

7

7

7

7

overall Heat transfer coefficient, fouling factors, Shell & tube heat exchangers, LMTD correction factors, Extended surface heat exchangers, Fin efficiency and fin effectiveness.

Text Books

"A Heat Transfer Textbook", John H. Lienhard, Phlogiston Press, Cambridge, Massachusetts, U.S.A. 4th Edition.

Reference:

Books

- 1. "Heat Transmission", W. H. McAdams, McGraw Hill, 3rd Edition.
- 2. "Process Heat Transfer", D. Q. Kern, McGraw Hill, 1st Edition.
- 3. "Unit Operations of Chemical Engineering", McCabe W L, Smith J C, Harriott P, 7th Ed. McGraw Hill, 2005.
- 4. "Heat Transfer", J. P. Holman, McGraw Hill, Tenth Edition.
- 5. "Perry's Chemical Engineer's Handbook", Don W. Green, Robert H. Perry, McGraw-Hill: New York, 8th Edition.
- 6. "Handbook of Heat Transfer", Warren M. Rohsenow , James P. Hartnett, Young I. Cho, McGraw-Hill Education: New York,1998

Class & Semester	:	S. Y.	S. Y. B. Tech (Food Technology), Part II, Semester III								
C Tid	I	TT 4.70			• .			G G I		777704 47	
Course Title	:	Heat T	ran	ster La	boratory	7		Course Code:	:	FT214L	
Teaching Scheme (Hours)	:	2 hr /w	eek:	= 2 x13	= 26 hou	rs	Credits		: 1		
Evaluation Scheme (Marks)	:	IPE IOE	:	50 Nil	EPE EOE	:	Nil Nil	Duration of Exam (in case of External Evaluation)	:	02 hours	
Revision:	:	Third	•	•		•		Month	•	: Decem 2016	ber

Laboratory work in Engineering Physics, Chemistry-I and Fluid

Pre-requisites Flow Operations.

Type of Course : Practical

Core Course Domain

: Cognitive: Understand, Apply, Analyze, Evaluate, Create

Affective: Awareness, Respond, Value, Organize

Skills Imbibed Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 1. To allow budding food process engineers to understand fundamentals of some major heat transfer operations through experimental set ups.
- 2. To provide them hands on training on important heat transfer devices and to motivate them for team work.
- 3. To provide them exposure in the laboratory like a miniature process plant environment using steam and the other process utilities.
- 4. To develop skill of learners in safe handling of major heat transfer equipment/devices, in close observation of their operation, in developing analytical ability in correlating the performance of the devices with their operational conditions.

Course Outcomes:

- 1. Understanding fundamentals of some major Heat transfer operations through practical work.
- 2. Getting set for practical heat transfer operations.
- 3. Application of heat transfer design principles and be mechanic for heat transfer devices.
- 4. Building foundation for process intensification and be adapted to handle heat transfer operations.
- 5. Learning to work in group and imitate Standard Operating Procedure for practical work v.i.z. responding to the guide.
- 6. Motivation towards innovations for novel systems of heat transfer.

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Practical List

- 1. **Metal Rod Apparatus:** To understand conduction heat transfer.
- 2. **Composite Wall Apparatus:** To understand conduction heat transfer through composite system.
- 3. **Sphere in Sphere Apparatus:** To study heat flow through a sphere and to estimate thermal conducting of powdered insulating material using the set up.
- 4. Thermal Conductivity of liquid Apparatus: To estimate thermal conductivity of liquid.
- 5. **Heat Transfer through lagged pipe Apparatus:** To analyze problems involving steady state heat conduction in simple geometries with lagged material
- 6. **Unsteady state heat transfer Apparatus:** To estimate the film heat transfer coefficient between the medium in which body is heated.
- 7. **Heat Transfer through Agitated vessels Apparatus:** To understand heat transfer during agitation and mixing.
- 8. **Natural Convection Apparatus:** To understand fundamentals of convective heat transfer process and to evaluate heat transfer coefficients for natural convection.
- 9. **Forced Convection Apparatus:** To understand fundamentals of convective heat transfer process and to evaluate heat transfer coefficients for forced convection.
- 10. **Stefan Boltzmann's Apparatus:** To understand radiation heat transfer through verification of the basic law of radiation.
- 11. **Emissivity Measurement Apparatus:** To understand radiation heat transfer and to evaluate emissivity of a material.
- 12. Critical Heat Flux apparatus: To study different types boiling and to draw boiling curve.
- 13. **Study of Heat Exchangers:** Analyze heat exchanger performance of different types of heat exchangers by using the method of log mean temperature difference or by method of heat exchanger effectiveness.
 - i. Shell & Tube Heat Exchanger ii. Double pipe Heat Exchanger iii. Finned tube heat Exchanger
 - iv. Plate Type Heat Exchanger v. Fluidized Bed Heat Transfer Unit

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference:

Books

- 1. "Process Heat Transfer", D. Q. Kern, McGraw-Hill, 1st Edition.
- 2. "Heat Transfer: Principles and Applications", Binay K. Dutta, Prentice-Hall of India, 1st Edition.
- 3. "Heat Transmission", William H. Mc Adams, McGraw-Hill, 3rd Edition.

Class & Semester	:	S. Y. B.Tec	ch (Food Tech	no	logy) Part I	I, Semester III		
Course Title	:	Programmi Technologis	ing practices	for	Course Code:	:	FT215L	
Teaching Scheme (Hours)	:	minimum Tutorial= Practical	1 hours/weeks= 1 x 13 weeks= 13 hrs minimum Tutorial= Nil Practical 2 hours/weeks= 2 x 13 weeks= 26 hrs					01+ 00 +01 =02
Evaluation Scheme (Marks)	:	CIE = Nil SEE = Nil	IPE=50 IOE=Nil EPE=Nil	:	Grand Total=50	Duration of IPE	:	2 hours
Revision:	:	Third		1	1	Month	:	December 2016

Pre-requisites	:	In order to complete the course studies successfully, it is important to have a good command of English. The learners ought to be sound in the basic concepts of basic computer programmes
Type of Course	:	Practical
Course Domain	:	Core
Skills Imbibed	:	Cognitive: Recall, Understand, Apply, Affective: Awareness Psychomotor: manipulation

Course Assessment Methods:

Student is evaluated during Internal Practical Examination.

Course Objectives:

- 1. To Identify and describe the purpose of various components of the VB integrated development environment (IDE).
- 2. To understand and apply Arithmetic operators, Logical operators and relational operators to build small application using Visual Basic.
- 3. To understand the concept of strings and their use in built-in functions.
- 4. To construct conditional and repetition statements and other control structures.
- 5. To Create Windows applications that store and access data to and from files and databases.

- 1. Students will be able to Design, create, build, and debug Visual Basic applications.
- 2. Students will be able to Explore Visual Basic's Integrated Development environment (IDE).
- 3. Students will be able to implement syntax rules in Visual Basic programs.
- 4. Explain variables and data types used in program development.
- 5. Apply arithmetic operations for displaying numeric output.
- 6. Apply procedures, sub-procedures, and functions to create manageable code.

Curriculum Content	Hours
Unit: I Introduction to Visual Basic Programming	
Menu bar, Tools bar, Project explorer, Tool box, Properties windows, Form Code, Controls: Command Buttons, Label, Textbox, Pointers, Picture box, frame, Naming Controls, Properties for controls: Height, Width, Left, Top, Font, Forecolor, Backcolor, Name, Caption, Text, and Visible, Events, Saving Visual Basic Project, Examples: Food Engineering Applications	02
Unit:II Mathematics Arithmetic Operations: +, , *, /, mod, ^. (Using Simple Example for each Operation, Logical Operations. AND, OR, NOT. And the Truth Table for each Operation, (Using Simple Example for each Operation), Relational Operation: >, <, >=, <=, <>, =, String	03
Concatenation (&),Operation Precedence. For all arithmetic, logical, relational operators, Print statement and Formatting, Illustrate (colon, comma, and semicolon), Examples: Food Engineering Applications	
Unit:III Built in Functions Builtin math functions: Abs(x), Int(x), Rnd(x), sgn(x), sqr(x), str(x), val(x), round(x, n), CInt(x), Fix(x),String Functions, InputBox, MsgBox, Examples: Food Engineering Applications.	02

II:4-IV Coloo4: o C404	/ Domestation Characteries										
	Unit:IV Selection Structure/ Reputation Structure Single Selection: If/Then structure, Double Selection: If/Then/Else structure, Nested										
		02									
	elect Case Multiple Selection Structure, For Next Loop,	02									
	hile Loop, Do Loop Until, Exit Do, Exit For, Examples:										
	Food Engineering Applications Unit: V Variables										
Data Types: Boolean, Integer, Long, Single, Double, String, Valid Naming of Variables, Initial Value for each Type of the Variables (Initial Value for each Data											
The state of the s	· · · · · · · · · · · · · · · · · · ·										
	ariable Type in Bytes, How to Declare Variables. (Dim variablename As Data type, Using Suffix: Integer, Long,										
	onstant Variable, Examples: Food Engineering Applications.	02									
Single, Double, String, Co	onstant variable, Examples. Food Engineering Applications.										
Unit: VI Database											
	a Control DAO, ADO, RDO, Visual Basic and Access										
Connectivity, Examples: I	Food Engineering Applications	02									
Practicals:											
1.To study VB environme											
	form to perform mathematical operations.										
3. To study date, string an											
_	ng image control and scroll bars.										
5. To design menu editor	as text editor.										
6. To design stop watch.											
7. To design form using fi											
8. To design form using A											
9-13 To build a small Foo	od Engineering applications.										
T (D 1	:										
Text Books	•										
1. "Programming w	rith Microsoft Visual Basic 2012", Diane Zak, Course 7	Technology,									
Cengage Learning	g, 6 th edition.										
Reference Books											
1) "Mastering M	licrosoft Visual Basic 2008", Evangelos Petroutsos, Wiley										
publishing,Inc., 1 st edition.											
2) "Visual Basic 6: The Complete Reference", Noel Jerke, Osborne/McGraw-Hill											
,											
3) E-Book: "Microsoft Visual Basic 2010", Michael Halvorson											

Class & Semester	:	S	S. Y. B.Tech (Food Technology)Part II, Semester III							
Course Title	:	Environmen	tal Studies			Course Code:	:	HS211		
Teaching Scheme (Hours)	:	Lectures 2 hours/wee hours Tutorial= 00 Practical= 0) hour/we	ek		Total Credits	:	Nil		
Evaluation Scheme (Marks)	:	CIE = 00 SEE = 70	IPE=30	:	Grand Total=100	Duration of SEE	:	2 hours (SEE at the yearend)		
Revision:	:	Third				Month	:	December 2016		

Pre-requisites	:	Engineering Chemistry
Type of Course	:	Theory and field work
Course Domain	:	Humanities and Applied Science
Skills Imbibed	:	Affective : Awareness, Respond, Value, Organize Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Project / Field work
- 2. Semester End Examination.

Course Objectives:

- 1. To recall fundamental physical and biological principles those govern natural processes.
- 2. To understand the importance of ecological balance for sustainable development.
- 3. To Understanding the impacts of developmental activities and mitigation measures and to further understand the environmental policies and regulations.
- 4. To identify the complex relationships between scientific approaches to environmental issues and political, social, economic, and ethical perspectives on the environment.
- 5. To collect and interpret scientific data in both field and laboratory settings.
- 6. To integrate and apply perspectives from across the natural sciences, social sciences, and the humanities in the context of complex environmental problems.
- 7. To communicate scientific information to both professional and lay audiences.

- 1. Develop an understanding of different natural resources including renewable resources.
- 2. Realize the importance of ecosystem and biodiversity for maintaining ecological balance.
- 3. Aware of important acts and laws in respect of environment.

- 4. Demonstrate critical thinking skills in relation to environmental affairs
- 5. Develop an understanding of environmental pollutions and hazards due to engineering/technological activities and general measures to control them.
- 6. Demonstrate knowledge and application of communication skills and the ability to write effectively in a variety of environmental contexts.
- 7. Demonstrate an ability to integrate the many disciplines and fields that intersect with environmental concerns.
- 8. Demonstrate an appreciation for need for sustainable development and role of science.

Curriculum Content	Hours
UNIT I: Significance of environmental studies	04
Multidisciplinary nature of environmental studies Need for public awareness.	
a) Forest resources: Use and over-exploitation, deforestation, Timber extraction, mining, dams and their	
effects on forests and tribal people. b) Water resources: Use and over-utilization of surface and ground	
water, floods, drought, conflicts over water, dams-benefits and problems. c) Mineral resources: Usage and	
exploitation, environmental effects of extracting and using mineral resources. d) Food resources: World	
food problem, changes caused by agriculture effects of modern agriculture, fertilizer-pesticide problems.	
e) Energy resources: Growing energy needs, renewable and non-renewable energy sources, use of	
alternate energy sources. f) Land resources: Land as a resource, land degradation, man induced landslides,	
soil erosion and desertification. g) Role of an individual in conservation of natural resources. h) Equitable	
use of resources for sustainable lifestyle.	
UNIT II: Ecosystems	
Concept of an ecosystem, Structure and function of an ecosystem, Producers, consumers and	04
decomposers, Energy flow in the ecosystem, Ecological succession, Food chains, food webs and ecological	04
pyramids. Introduction, types, characteristics features, structure and function of the following Ecosystem:	
- a) Forest ecosystem, b) Grassland ecosystem, c) Desert ecosystem,	
d) Aquatic ecosystems (ponds, streams, lakes, rivers, oceans, estuaries)	
UNIT III: Biodiversity and its Conservation	
Introduction – Definition: genetic, species and ecosystem diversity, Biogeographical classification of India,	04
Value of biodiversity: consumptive use, productive use, social, ethical, aesthetic and option values.;	
Biodiversity at global, National and local levels.; India as a mega-diversity nation; Western Ghats as a biodiversity region; Hot-spots of biodiversity; Threats to biodiversity: habitat loss, poaching of wildlife, man-	
wildlife conflicts; Endangered and endemic species of India; Conservation of biodiversity: In-situ and Ex-	
situ conservation of biodiversity.	
UNIT IV: Environmental Pollution	
Definition: Causes, effects and control measures of:	04
a) Air pollution, b) Water pollution, c) Soil pollution, d) Marine pollution, e) Noise pollution, f) Thermal	
pollution, g) Nuclear hazards	
Solid waste Management: Causes, effects and control measures of urban and industrial wastes.	
• Role of an individual in prevention of pollution. • Pollution case studies • Disaster management: Floods,	
earthquake, cyclone and landslides. Tsunami UNIT V: Social Issues and the Environment	
From Unsustainable to Sustainable development; Urban problems related to energy; Water conservation,	05
rain water harvesting, watershed management; Resettlement and rehabilitation of people; its problems	UO
and concerns; Environmental ethics: Issue and possible solutions; Climate change, Global warming, acid	
rain, ozone layer depletion, nuclear accidents and holocaust; Wasteland reclamation; Consumerism and	

waste products.

UNIT VI: Environmental Protection

Environment Protection Act.; Air (Prevention and Control of Pollution) Act.; Water (Prevention and control of Pollution) Act; Wildlife Protection Act; Forest Conservation Act; Population Growth and Human Health, Human Rights. ;Field Work--Visit to a local area to document environmental assets river/forest/grassland/hill/mountain or Visit to a local polluted site –urban/rural/Industrial/Agricultural or Study of common plants, insects, birds or Study of simple ecosystems-ponds, river, hill slopes, etc.

05

Text Books

:

- 1. Agarwal, K. C. 2001, Environmental Biology, Nidi Publ. Ltd., Bikaner.
- 2. Bharucha Erach, The Biodiversity of India, Mapin Publishing Pvt. Ltd., Ahmedabad, 380013, India, Email:mapin@icenet.net (R)
- 3. Brunner R. C., 1989, Hazardous Waste Incineration, McGraw Hill Inc. 480p

Reference Books

:

- 1. Clark R. S., Marine Pollution, Clanderson Press Oxford (TB) Pg No. 6
- 2. Cunningham, W. P. Cooper, T. H. Gorhani, E. & Hepworth, M. T. 2001, Environmental Encyclopedia, Jaico Publ. House, Mumbai, 1196p
- 3. De A. K., Environmental Chemistry, Wiley Eastern Ltd.
- 4. Down to Earth, Centre for Science and Environment (R)
- 5. Gleick, H., 1993, Water in crisis, Pacific Institute for Studies in Dev., Environment & Security. Stockholm Env. Institute. Oxford Univ. Press 473p
- 6. Hawkins R. e., Encyclopedia of Indian Natural History, Bombay Natural History Society, Bombay (R)
- 7. Heywood, V. H. & Watson, R. T. 1995, Global Biodiversity Assessment, Cambridge Univ. Press 1140p.
- 8. Jadhav, H. & Bhosale, V. M. 1995, Environmental Protection and Laws, Himalaya Pub. House, Delhi 284p.
- 9. Mckinney, M. L. & Schocl. R. M. 1996, Environmental Science Systems & Solutions, Web enhanced edition
- 10. Mhskar A. K., Matter Hazardous, Techno-Science Publications (TB)
- 11. Miller T. G. Jr., Environmental Science, Wadsworth Publishing Co. (TB)
- 12. Odum, E. P. 1971, Fundamentals of Ecology, W. B. Saunders Co. USA, 574p.
- 13. Rao M. N. & Datta, A. K. 1987, Waste Water Treatment, Oxford & IBH Publ. Co. Pvt. Ltd.,
- 14. Sharma B. K., 2001, Environmental Chemistry, Goel Publ. House, Meerut
- 15. Survey of the Environment, The Hindu (M)
- 16. Townsend C., Harper, J. and Michael Begon, Essentials of Ecology, Blackwell Science (TB)
- 17. Trivedi R. K., Handbook of Environmental Laws, Rules, Guidelines, Compliances and Standards, Vol. I and II, Enviro Media (R)
- 18. Trivedi R. K. and P. K. Goel, Introduction to air pollution Techno-Science Publications (TB)
- 19. Wagner K. D., 1998, Environmental Management, W. B. Saunders Co. Philadelphia, USA.
- (M) Magazine
- (R) Reference
- (TB) Textbook
- 20. Paryavaram Swhastra Gholap T. N.
- 21. Paryavaram Shastra Gharapure.
- 22. Paryavaran Vighyan V. R. Ahirrao Nirali Prakashan, Pune.
- 23. Paryavaram Shastra Parichay Jay Kumar Magar Vidya Prakashan, Nagpur.
- 24. Desh Ka Paryavaran Anupam Misra, Ganolai santi Pratisthan. New Delhi.

	:	S. Y. B. Tech (Food Technology), Part II, Semester III										
Class & Semester												
Course Title	:	Soft Skills Deve	oft Skills Development									
Teaching Scheme (Hours)	:	2 hr /week= 2 x1	3=	26 h	ours		Credits	:	Nil			
Evaluation Scheme (Marks)	:	Assignments Viva voce								Not Applicable		
Revision:	:	Third				Month	:	December 2016				

Pre-requisites	H.S.C level English Language Competency	
Type of Course	Audit Course at institute level	
Course Domain	Humanity and Arts	
Skills Imbibed	Cognitive: Understand, Predicting Situation, Comprehend, Affective: Receive, Listen, Respond, Showing self reliance, Organize Psychomotor: Imitation, adaptation, articulation, origination	

Course Assessment Methods:

The students will be given five assignments each for 10 marks. At the end of the course, there will be a written test of 25 marks and a viva voce of 25 marks. There will be assessment for a total of 100 marks. Based on the marks obtained, they will be awarded with a grade similar to other credit courses. Though it is an audit course, obtaining passing grade is essential.

Course Objectives:

- 1. To develop effective communication skills (spoken and written).
- 2. To develop effective presentation skills.
- 3. To compete successfully in the business environment.
- 4. To generate ability in the learners to put their domain knowledge into effective practice.
- 5. To make the students self-confident individuals by mastering inter-personal skills, team management skills, and leadership skills.
- 6. To prepare the learners to take part effectively in various selection procedures adopted by the recruiters and to increase employment opportunities

- 1. Students are able to expertise in self development, effective communication skills and interview skills
- 2. Understand how to handle situation and take decision
- 3. Equip to any sort of interviews particularly job interviews
- 4. Acquaintance with documentation skills
- 5. Become self reliant and responsible

	Curriculum Content	Hours				
Unit I : Self Develo	pment	02				
Self analysis, creativ	vity, attitude, motivation, goal setting. Importance of career visioning and planning.					
Unit II : Effective	Communication Skills	06				
Importance of com	munication, Communication process, Elements of communication, Communication Types-					
verbal and non verbal, objectives of communication. Business Communication, current English usage, debates,						
language games, situational dialogues, precise writing, essay writing, presentations.						
Unit III : Behavior		08				
	ts: Aptitude and personality assessment, suggestions for improvement, Team Skills: Team ship, evolution of groups into teams, group dynamics, emergence of leadership, intra-group					
Management: Pare effective time man	up dynamics, conflict management, inter dependency, assessment of team-based projects, Time to's Principle, Parkinson's Laws, Murphy's Laws, Law of Clutter, prioritization, goal setting, agement, Interpersonal Skills: Negotiations, listening skills, social skills, assertive skills, nunications, Leadership Skills: Concepts of leadership, leadership styles, insights from great					
Unit II : Document	ation	03				
	nal report, study tour report, project report, Writing proposal-solicited proposals and unsolicited	05				
proposals.	mi report, study tour report, project report, writing proposur solicited proposurs and unsolicited					
Unit III: Emotiona	l Intelligence	04				
	Nature of emotional intelligence, emotional intelligence applied windows of opportunity,					
emotional literacy.						
Unit VI: Interview	Skills	03				
Importance of Inter actual career planning	view Skills, Resume Building, Group discussion and personal interview, Psychometric Test, ng.					
Text Book	:					
1. Soft Skills, 2015,	Career Development Centre, Green Pearl Publications.					
Reference Books	:					
1. "Seven Habits of	Highly Effective Teens", Covey Sean, , New York, Fireside Publishers, 1998.					
2. "How to win Frie	nds and Influence People", Carnegie Dale, New York: Simon & Schuster, 1998.					
3. "I am ok, You are	ok ",Thomas A Harris, New York-Harper and Row, 1972					
	igence", Daniel Goleman, Bantam Book, 2006					
5 "Effective commi	unication skill", MTD training & Ventus publishing ApS ISBN 978-87-7681-598-1.					

Class & Semester	:	S. Y. B. Te	S. Y. B. Tech (Food Technology) Part II, Semester IV						
Course Title	:	Food Proce	ess Engineer	ring	g-I	Course Code:	:	FT221	
Teaching Scheme (Hours)	:	hours minitutorial =0 =1x13 week	Lectures 3 hours/weeks=3 x 13 weeks= 39 hours minimum Tutorial =01 hours/week =1x13 weeks=13 hours minimum Practical= 02 hours/week				:	04+01 =05	
Evaluation Scheme (Marks)	:	CIE =50 SEE = 50	IPE=Nil IOE=50 EOE= 50	:	Grand Total=200	Duration of SEE	:	3 hours	
Revision:	:	Third				Month	:	December 2016	

Pre-requisites	:	In order to complete the course studies successfully, it is important to have a good knowledge of applied mathematics and basic science
		knowledge of applied mathematics and basic science
Type of Course	:	Theory
Course Domain	:	Core
	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate
Skills Imbibed		Affective : Awareness, Respond, Value, Organize
		Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

- 1. To understand units and dimensions, ability to solve engineering problems related to food processing, and familiarization with some food processing unit operations.
- 2. To understand the principle involved in food processing engineering
- 3. To enable the students to solve problems in food processing industries
- 4. To impart knowledge of the principles of unit operations and its importance in food Industries.
- 5. To understand the principle and working of various processing equipments
- 6. To know the methods of product recovery

Course Outcomes:

- 1. Understanding of all unit operations and its applications in food processing
- 2. Calculate the effectiveness and the amount of heat exchanged in heat exchangers under various flow conditions and determination of the ideal conditions for heat transfer
- 3. Calculate the amount of mass going into the system and coming out of the system for the processing of products
- 4. Knowledge of the food processing equipments used for the different unit operations.
- 5. Describe how various unit operations work individually and together.
- 6. Understanding of the principles in solving the problems of food processing industries

Unit I: Introduction and Size Reduction	Hrs						
Introduction: Unit operations in food engineering, Mass and energy balance Size Reduction: Particle size analysis, equipments, application to food industries							
Unit II: Mixing Basic theory of solid – solid mixing, liquid- liquid mixing, equipments, applications to food industries.							
Unit III: Filtration and Centrifugation Filtration: Theory of filtration, industrial filters, applications to food industries Centrifugation: Theory of centrifugation, equipments, applications for food industries							
Unit IV: Distillation Vapour liquid equilibria, batch and continuous distillation, steam distillation, equipments and applications to food industries							
Unit V: Diffusion and Gas Absorption Diffusion-Basics of absorption- Diffusion of gases in liquid and solid foods. Gas Absorption – Principles, equipments and applications in food processing							
Unit VI: Extraction Solvent extraction, leaching, equipments. applications for food industries							
Text Books :							
 D G Rao 2012. Fundamental of Food Enginnering. PHI Learning Private Limited, New Delhi. George D. Saravacos and Athanasios E. Kostaropoulos. 2002. Handbook of Food Processing Equipment. Springer Science+Business Media, New York, USA 							
Reference Books :							
1. McCabe and Smith "Unit Operations" McGraw-Hill, New York							
2. Transport Process and Unit Operations: Geankoplis, 3rd Edn. PHI							

3. Singh and Heldman "Introduction to food engineering" Academic Press publications

Class & Semester	:	S. Y. I	S. Y. B. Tech (Food Technology), Part II, Semester IV								
Course Title	:		ood Process Engineering-I aboratory Course Code: : FT221L								
Teaching Scheme (Hours)	:	2 hr /w	hr /week= 2 x13= 26 hours					Credits	:	1	
Evaluation Scheme (Marks)	:	IPE IOE	:	Nil 50	EPE EOE	:	Nil 50	Duration of Exam (in case of External Evaluation)	:	02 hours	
Revision:	:	Third						Month	:	December 2016	

Pre-requisites: Laboratory work in Engineering Physics and Engineering Chemistry.

Type of Course : Practical

Course Domain : Core

: Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 1. To study the mass balance and energy balance of food processing.
- 2. To study the properties and pretreatment of food materials.
- 3. To study the unit operations of food processing
- 4. To have the practical experience of all unit operations

- 1. Understanding the calculations of mass balance and energy balance of food processes
- 2. Understanding of basic principles of unit operations
- 3. Understanding of practical experience of various equipments
- **4.** Reorganization of key food processing and product parameters that can influence microbiological safety and quality of the processed product.

Practical List

- 1. Design problems on mass balance equations
- 2. Design problems on energy balance equations
- 3. Experiment on Particle size analysis
- 4. Experiment on Size reduction
- 5. Experiment on Solid Solid Mixing
- 6. Experiment on liquid Liquid mixing
- 7. Experiment on Filtration
- 8. Experiment on Centrifugation
- 9. Experiment on sieve tray Distillation
- 10. Experiment on Diffusion
- 11. Experiment on Gas Absorption
- 12. Experiment on Solvent extraction
- 13. Experiment on Leaching

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. "McCabe and Smith "Unit Operations" McGraw-Hill, New York
- 2. Transport Process and Unit Operations: Geankoplis, 3rd Edn. PHI
- 3. Singh and Heldman "Introduction to food engineering" Academic Press publications

Class & Semester	:	S. Y. B. Tech (Food Technology)	Part II, Seme	est	er IV
C T'd					ET222
Course Title	:	Food Additives and Contaminants	Course Code:	:	FT222
Teaching Scheme (Hours)	:	Lectures 4 hours/weeks=4 x 13 weeks= 52 hours minimum Tutorial= 0 hour/week Practical= 02 hours/week	Total Credits	:	04+01=05
Evaluation Scheme (Marks)	:	CIE = 50	Duration of SEE		3 hours
Revision:	:	Third	Month	:	December 2016

	•	In order to complete the course studies successfully, it is important to have a good
Pre-requisites	'	command of English. Other Pre-requisites include basic knowledge of Food science and
_		food analysis, labeling and food microbiology.
Type of Course	:	Theory
Course Domain	:	Core
	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate
Skills Imbibed		Affective : Awareness, Respond, Value, Organize
		Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II I, home assignments
- 2. Semester End Examination.

Course Objectives:

- 1. To understand the permitted levels of all types of food additives.
- 2. To understand the principles of chemical preservation of foods.
- 3. To maintain the nutritional quality of food, to enhance stability with resulting reduction in waste, to make food more attractive.
- 4. To provide efficient aids in processing, packaging and transport.
- 5. To demonstrate skills for the identification and control of microbial and other food contaminants.
- 6. To provide students an in depth understanding of how the science of food additives and contaminants, Toxicology is applied to food and feed safety including food regulation and risk assessment.

- 1. Better understanding of the functions of different food additives in improving shelf life, quality, texture and other physical and sensory characteristics of foods.
- 2. Exposure about food additives related to bakery and confectionary.
- 3. Exposure about types and chemical properties of pigments and processing effect.
- 4. Exposure about types and chemistry of flavoring compounds.
- 5. Exposure about sensory analysis.
- 6. Provide students with a basic understanding of the principles of food toxicology.

	Hours		
UNIT-I Food Additives General attributes, buffer system/salt/, acids, antimicrobial agents, chelating agent and sequestrants,	12		
UNIT- II Bakery and Confectionery Additives Antioxidants, emulsifiers, non nutritive and low calories sweeteners, stabilizer and thickeners and dough improvers.	8		
UNIT-III Flavour Philosophy and definition of flavor, chemistry of flavoring compounds.	8		
UNIT-IV Pigments Chlorophyll, carotenoid, phenolic & flavonoids, Betalins effect of processing on pigments, Retention of natural food colours, Food colourants, Regulatory aspects.			
UNIT-V Sensory Assessment Sensory assessment of flavours ,technology for flavour retention.			
UNIT-VI Food Toxicology Inherent toxicants terms in toxicology, safety evaluation using traditional & modern approach contaminants, pesticide residues, toxicology & public health.	8		
Reference Books :			

- 1. Furia, T. E., Handbook of Food Additives, CRC Press, Boca Raton, Flor., 1980
- 2. Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker
- 3. Fennema O.R. "Food Chemistry" François Castaigne, Rohanie Maharaj 1996
- 4. DeMan "Food chemistry" Hardcover, A V I Publishing Company, 1982

Class & Semester	:	S. Y. B	S. Y. B. Tech (Food Technology), Part II, Semester IV							
	ı								ı	
Course Title	:		ood Additives and Contaminants aboratory					Course Code:	:	FT222L
Teaching Scheme (Hours)	:	2 hr /wee	2 hr /week= 2 x13= 26 hours					Credits	:	1
Evaluation Scheme (Marks)	:			,	EPE EOE	:	50 Nil	Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third						Month	:	December 2016

Pre-requisites

• In order to complete the course studies successfully, it is important to have a good command of English. Other Pre-requisites include basic knowledge of Food science and food Analysis, Labeling and Food Microbiology.

Type of Course

Practical

Course Domain

Core

Cognitive: Understand, Apply, Analyze, Evaluate, Create

Affective: Awareness, Respond, Value, Organize

Skills Imbibed

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

On completion of the course the students are expected to

- 1. Be able to study types of food adulterant used and their detection.
- 2. Be able to study different types of pigments, coloring compounds
- 3. Be able to determine the major macro and micronutrients
- 4. Be able to recognize the common analytical techniques for detection of food toxicants
- 5. Be able to demonstrate critical and analytical thinking as well as problem solving skills

- 1. Recognize the common analytical techniques for detection of food adulterant
- 2. Detection of different types of vitamins, pigments.

- 3. Identification of appropriate techniques for analysis
- 4. Detection of different properties of protein, starch, fat
- 5. Understanding the effect of emulsifier, hydrocolloids as food additive.
- 6. Knowledge of regulations and the monitoring agencies involved in controlling the safer use of additives in foods.

Practical:

- 1. Determination of Food adulterants
- 2. Determination of carotenoids
- 3. Assessments of hydrocolloids as food additives
- 4. Estimation of tannins
- 5. Estimation of phytic acids
- 6. Solubility characteristics of starches.
- 7. Determination of Vitamin E
- 8. Effect of emulsifier in food products
- 9. Determination of chlorophyll
- 10. Determination of Vitamin C
- 11. Determination of color
- 12. Foaming properties of proteins
- 13. Determination of peroxide value of fat

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. Furia, T. E., Handbook of Food Additives, CRC Press, Boca Raton, Flor., 1980
- 2. Branen, A. F. et al (2001). Food Additive s, 2nd Edition, Marcel. Dekker
- 3. Fennema O.R. "Food Chemistry" François Castaigne, Rohanie Maharaj 1996
- 4. DeMan "Food chemistry" Hardcover, A V I Publishing Company, 1982

Class & Semester	:	S. Y. B.Tech (Food Technology) Part II, Semester IV						
	1							
Course Title	:	FOOD BIOCHEMISTRY	Course Code:	:	FT223			
Teaching Scheme (Hours)	:	Lectures 4 hours/weeks= 4 x 13 weeks= 52 hours minimum Tutorial= 00 hour/week Practical= 02 hours/week	Total Credits	:	04+01 =05			
Evaluation Scheme (Marks)	:	CIE = 50 IPE=Nil : Grand : Grand Total=150	Duration of SEE	:	3 hours			
Revision:	:	Third	Month	:	December 2016			

Pre-requisites		In order to complete the course studies successfully, it is important to have a good
		command of English.
Type of Course	:	Theory
Course Domain	:	Core
	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate
Skills Imbibed		Affective : Awareness, Respond, Value, Organize
		Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

To enable the students to understand

- 1. Biochemical cycles
- 2. Biosynthesis of macromolecules and micromolecules
- 3. The changes of biological function of food components due to food processing and storage
- 4. Understand the major chemical reactions that limit shelf life of foods
- 5. life processes at the molecular level
- 6. The basic concepts in food chemistry and food analysis.

Course Outcomes:

Upon successful completion of this course, students will be able to:

- 1. Understand the chemistry underlying the properties and reactions of various food components
- 2. To describe the biochemistry process, basic concept of nutrients and the relationship of the consumption of foods to nutritional status and health
- 3. Identify some of common reaction mechanisms in biochemical processes
- 4. Describe how enzymes work and know how to determine basic enzyme kinetics
- 5. Analyze and predict how the composition and conditions within a food influence the functional properties of food molecules.
- 6. Comprehend the degree of knowledge that is being acquired through current research in understanding how cells function

UNIT-I Digestion, absorption and metabolism Digestion, absorption and metabolism of carbohydrate, fat and protein.	Hours 12
UNIT-II Enzymes Chemical nature and nomenclature, classification, sources and properties, mechanism of action, Coenzyme and prosthetic groups application.	8
UNIT-III Biochemical changes in meat Biochemical changes in meat and meat products during processing.	8
UNIT-IV Biochemical changes in fruits and vegetables Biochemical changes in fruits and vegetables during post harvest handling, processing and storage.	8
UNIT-V Biochemical changes in milk Biochemical changes in milk and milk products during handling, processing and storage.	8
UNIT-VI Hormones Hormones – related to metabolic process.	8
Text Books :	

Text Books

- 1. Cox, M.M. and Nelson, David L. Lehininger, "Principles of Biochemistry". 5thEdition. W.H.Freeman, 2008
- 2. Murray, Robert K. etal., "Harper's Illustrated Biochemistry" 27th Edition. McGraw-Hill, 2006.
- 3. Satyanarayanan, U. "Biochemistry" Books and Allied. 2005
- 4. Voet, D.J. and J.G.Voet "Principles of Biochemistry" 3rd Edition. John wiley, 2008.
- 5. S. Rastogi, S.C. "Biochemistry" 2nd Edition. Tata Mc Graw-Hill, 2003

: Reference Books

- Lehninger, "Biochemistry", CBS. *Publications*, 1993.
 Stryer, "Biochemistry" W H Freeman, New York. 1988.
- 3. Voet and Voet, "Principles of Biochemistry" John Wiley & Sons, Inc.. Edition

Class & Semester	:	S. Y. B. Tech (Food Technology), Part II, Semester IV								
Course Title	:		FOOD BIOCHEMISTRY LABORATORY			Course Code:	:	FT223L		
Teaching Scheme (Hours)	:	2 hr /w	2 hr /week= 2 x13= 26 hours				rs	Credits	:	1
Evaluation Scheme (Marks)	:	IPE IOE	:	Nil Nil	EPE EOE	:	50 Nil	Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third			•			Month	:	December 2016

Pre-requisites

: In order to complete the course studies successfully, it is important to have a good command on English and subject proficiency.

i i i i quistics

: Practical

Course Domain

Type of Course

: Core

: Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed

Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

1. To learn and understand the principles behind the qualitative and quantitative estimation of biomolecules (proteins, carbohydrates, lipids, metabolites etc.,) and laboratory analysis of the same in the body fluids.

- 1. The student will able to evaluate the food component on qualitative and quantitative basis.
- 2. Be able to know the laboratory techniques and different methodology for biological assay.
- 3. The student will be able to apply principles of laboratory techniques.
- 4. The student will be able to identify appropriate techniques for analysis.
- 5. Have sufficient knowledge of food biochemistry to control reactions in foods.
- 6. Develop a practical proficiency in a food biochemistry and analysis laboratory.

Practical List

- 1. Preparation of standard solutions and buffers
- 2. Quantitative determination of carbohydrate by DNSA method.
- 3. Isolation and characterization of starch
- 4. Quantitative determination of protein
- 5. Determination of vitamins C.
- 6. Determination of vitamins A.
- 7. Isolation of enzymes from various sources
- 8. Enzyme assay methods
- 9. Immobilization of enzymes.
- 10. Qualitative analysis of proteins
- 11. Qualitative analysis of lipids.
- 12. Quantitative analysis of lipids (Benedict's method etc.)
- 13. Quantitative estimation of blood glucose.

Lab Manual

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. Lehninger, "Biochemistry", CBS. Publications, 1993.
- 2. Stryer, "Biochemistry" W H Freeman, New York. 1988.
- 3. Voet and Voet, "Principles of Biochemistry" John Wiley & Sons, Inc.. Edition

Class & Semester	:	S. Y. B. Tech (Food Technology)	Part II, Semester IV			
Course Title	:	Human Nutrition	Course Code:	:	FT224	
Teaching Scheme (Hours)	:	Lectures 4 hours/weeks=4 x 13 weeks= 52 hours minimum Tutorial= 0 hour/week Practical= 02 hours/week	Total Credits	••	04+01 =05	
Evaluation Scheme (Marks)	:	CIE = 50	Duration of SEE	:	3 hours	
Revision:	:	Third	Month	:	December 2016	

Pre-requisites	:	Knowledge of basic food chemistry and food processing.
Type of Course	:	Theory
Course Domain	:	Core
Skills Imbibed	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate Affective: Awareness, Respond, Value, Organize Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

- 1. Be able to understand the physiological and metabolic functions of human digestive system
- 2. Be able to understand an overview of the major macro and micronutrients relevant to human health
- 3. Be able to explain how dietary recommendations are formulated
- 4. Be able to understand proper diet planning, nutritional facts for balanced nutrition and healthy diets
- 5. Be able to understand the role of diet in causing and preventing various diseases
- 6. Be able to get a basic foundation in human nutrition in preparation for any of the health professions.

Course Outcomes:									
Better understanding in physiological and metabolic functions of nutrients									
2. Familiarize nutritional assessment, RDA and Dietary Recommendations & guidelines.									
3. Understanding of energy and water balance									
4. Understanding of malnutrition ,their causes and nutrition education									
5. Effective understanding of Diet planning Principles, exchange lists, food labels and									
nutritional facts for balanced nutrition and healthy diets									
6. Understanding of Undesirable Constituents & toxic substances and their disorders.									
	Hours								
UNIT-I Introduction Scope concepts and importance of nutrition, human digestive system	110015								
Scope, concepts and importance of nutrition, human digestive system									
UNIT- II Nutritional aspects									
Nutritional aspects of carbohydrate, protein, lipids, water, vitamin and minerals, food, fad and									
faddism.	08								
UNIT- III Energy and water balance									
Energy and water balance, Water intake and losses, energy requirement, and physiological energy	10								
value, bomb colorimeter.									
UNIT- IV Malnutrition									
Types of malnutrition, multi-factorial causes, epidemiology of under nutrition and over nutrition,	10								
nutrition infection and immunity, nutrition education.									
UNIT- V Balance diet									
Balance diet, types of balance diet, diets for specific purposes.									
balance diet, types of balance diet, diets for specific purposes.	08								
UNIT- VI Undesirable Constituents & toxic substances									
Undesirable Constituents & toxic substances and their disorders, hormones.	08								
REFERENCE BOOKS:									
1. Shubhangini A. Joshi, "Nutrition and Dietetics" "Tata Mc Grow- Hill publishing									
Company Ltd,.New Delhi (1992)									
2. Dr. M. Swaminathan, Vol I & II "Foods and Nutrition" NIN Publications									
3. Swaminathan, N.Food Science and experimental foods, (1987) Ganesh Publications,									
Madras.									

Class & Semester	:	S. Y. B	S. Y. B. Tech (Food Technology), Part II, Semester IV						
Course Title	:	Human I	Human Nutrition Laboratory				Course Code:	:	FT224 L
Teaching Scheme (Hours)	:	2 hr /weel	2 hr /week= 2 x13= 26 hours			Credits	:	1	
Evaluation Scheme (Marks)	:	IPE :	Nil 50	EPE EOE	:	Nil Nil	Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third					Month	:	December 2016

Pre-requisites

Knowledge of basic chemistry and biology.

Type of Course

Practical

Course Domain

Core

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Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed

Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

- 2. Be able to understand the physiological and metabolic functions of nutrients.
- 3. Be able to determine the major macro and micronutrients relevant to human health
- 4. Be able to understand methods of nutritional assessment, RDA and guidelines.
- 5. Be able to explain how dietary recommendations are formulated
- 6. Be able to describe the methods used to carry out nutrition research
- 7. Be able to Compute energy requirement on the basis of physical activity

Course Outcomes:

- 1. Understanding of methods used to assess nutrition status
- 2. Familiarize nutritional assessment, RDA and Dietary Recommendations & guidelines.
- 3. Understanding and determining BMR and body surface area
- 4. understanding of food composition and energy balance in dietary planning
- 5. Effective understanding of diet plan formulation for health and for nutrition-related disorders.

6. Identifying appropriate techniques for Biochemical analysis of blood, urine

Practical:

- 1. Calculation of BMR and body surface area,
- 2. Calculation of energy value of food.
- 3. Preparation of balance diet,
- 4. Anthropometric measurements,
- 5. Biochemical analysis of blood
- 6. Biochemical analysis of urine.
- 7. Computation of energy requirement on the basis of physical activity
- 8. ACU units.
- 9 Quantitative determination of carbohydrate by DNSA method.
- 10. Determination of vitamins C.
- 11. Determination of vitamins A.
- 12. Qualitative analysis of proteins
- 13. Determination of Ca, Iron, phosphorus and Ash content of food

Lab Manual:

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

- 1. Steeter U.L, 'Fluid Mechanics' V ed. Mc graw Hill Book Co., International Edn.1971.
- 2. Richardson J.E. and Coulson J.M. Chemical Engineering. 3rd ed. Vol. 1 Pergamon Press 1985.
- 3. Miohell B.I. Fluid and Particle Mechanics Pergamon Press 1970.
- 4. Gupta S.K., Momentum Transfer Operations, Tata McGraw Hill, 1979

Class & Semester	:	S. Y. B. Tech (Food Technology)	Part II, Semester IV		
	•				
Course Title	:	Fluid Mechanics	Course Code:	:	FT225
Teaching Scheme (Hours)	:	Lectures 4 hours/weeks=4 x 13 weeks= 52 hours minimum Tutorial= 0 hour/week Practical= 02 hours/week	Total Credits	:	04+01 =05
Evaluation Scheme (Marks)	:	CIE = 50	Duration of SEE	:	3 hours
Revision:	:	Third	Month	:	December 2016

Pre-requisites	:	Fundamental knowledge of applied mathematics, science and engineering
Type of Course	••	Theory
Course Domain	••	Core
Skills Imbibed	:	Cognitive: Recall, Understand, Apply, Analyze, Synthesize, Evaluate Affective: Awareness, Respond, Value, Organize Psychomotor: Imitation, manipulation, articulation, naturalization

Course Assessment Methods:

- 1. Continuous Internal Evaluation: Unit Test I & Unit Test II, Regular Tutorial, home assignments
- 2. Semester End Examination.

Course Objectives:

To give the student

- 1. To give fundamental knowledge of fluid, its properties and behavior under various conditions of internal and external flows.
- 2. Understand the dynamics of fluid flows and the governing non-dimensional parameters,
- 3. Apply concepts of mass, momentum and energy conservation to flows,
- 4. Practice in the analytical formulation of fluid mechanics problems using Newton's Laws of motion and thermodynamics
- 5. To inculcate the importance of fluid flow measurement and its applications in Industries.
- 6. An exposure to practical applications, work on a small design project, and the writing of a technical report related to the design project.

Course Outcomes:

- 1. An understanding of fluid mechanics fundamentals, including concepts of mass and momentum conservation.
- 2. An ability to apply the Bernoulli equation to solve problems in fluid mechanics.
- 3. An ability to apply control volume analysis to problems in fluid mechanics.
- 4. An ability to use potential flow theory to solve problems in fluid mechanics.
- 5. An ability to perform dimensional analysis for problems in fluid mechanics.
- 6. A knowledge of laminar and turbulent boundary layer fundamentals.
- 7. An understanding of how fluid mechanics apply to biological, environmental, and micro-fluid systems

Unit I Unit systems and Introduction to Fluid statics

8

Physical quantities, S.I., CGS, FPS engineering units, Conversion of Units, Units and Equations, dimensional analysis, Application of dimensional analysis, Problems. Nature of fluids, Hydrostatic equilibrium, Barometric equation, Hydrostatic equilibrium in centrifugal field, Manometers, Example, U tube, inclined tube manometers.

Unit II Fluid flow phenomena and Basic equations of fluids flow

12

Behaviour of flowing fluid, Types of flow, Newtonian and non-Newtonian Fluids, viscosity andmomentum flux, viscosities of gases and liquids, Turbulence, Reynolds experiment, Eddy viscosity, Flow in boundary layers, Laminar and Turbulent flow in Boundary layers, Boundary layer formation instraight tubes, Boundary layer separation and wake formation .Mass balance, mass velocity, momentum balance, and Bernoulli's equation without and withfriction, kinetic energy correction factor, correction for fluid friction, Pump Bernoulli's equation, Euler's equation, Problems.

Unit III Flow of incompressible fluids in conduits and thin layers

8

Shear stress distribution in a cylindrical tube, relation between skin friction and wall shear, thefriction factor. Relations between skin friction parameters .Laminar flow in pipes, Laminar flow ofNewtonian fluids. Average velocity, kinetic energy correction factor (Derivation), Momentum correction factor (Derivation), Hagen-poiseuille equation. Turbulent flow in pipes and closed channels. Velocity distribution for turbulent flow, universal velocity distribution equations for laminar sub layer and buffer layer, Relations between maximum and average velocities, Effect of roughness, the friction factor chart (Moody's diagram), friction factor in flow through channels of non-circular section, friction from changes in velocity or direction, Effect of

fittings and valves, couette flow, Layer flow with free surfaces, Flow through annulus, Problems.						
Unit IV Transportation and metering of fluids						
Pipe and tubing, joints and fittings. Prevention of leakage around moving parts. Valves like	8					
Gatevalve, globe valve, checks valve butterfly valve, needle valve, ball valve etc. Measurement of						
flowing fluids. Venturimeter, orifice meter, pitot tube, Rota meters, target meters, vortex-						
shedding meters ,turbine meters, positive displacement meters, magnetic meters: ultrasonic						
meters.						
Unit V Flow past immersed bodies Drag coefficients of typical shapes, form drag and stream lining, Friction in flow through bedsof	8					
solids, Erguns equation, Kozeny- Carman equation, Burke Plummer equation, Fluidization,						
Mechanism of fluidization, particulate and aggregative fluidization, minimum fluidization						
velocity, expansion of -fluidized beds, application of fluidization.						
Unit VI Flow of compressible fluids and Agitation of fluids						
Mach number, continuity equation, Total energy Balance, velocity of sound, ideal gas	8					
equations, the asterisk condition, stagnation temperature. Agitation of liquids, Agitation	8					
equipment, flow patterns in agitated vessels, circulation rates, Flow numbers, power consumption,						
power correlations, power correlations for specific impellers, effect						
of system geometry and calculations for power consumption						
TEXT BOOK: 1. Mc Cabe W.L. and Smith J.C, 'Unit operations of Chemical Engineer VII edition, Mcgraw Hill Book Co., International ed. 1993	ering.'					
Reference Books Refere						

Class & Semester	:	S. Y. B. Tech (Food Technology), Part II, Semester IV							
Course Title	:	Fluid Me	Fluid Mechanics Laboratory				Course Code:	:	FT225L
Teaching Scheme (Hours)	:	2 hr /week	2 hr /week= 2 x13= 26 hours			Credits	:	1	
Evaluation Scheme (Marks)	:	IPE : IOE :	Nil 50	EPE EOE	:	Nil Nil	Duration of Exam (in case of External Evaluation)	:	02 hours
Revision:	:	Third	Third			Month	:	December 2016	

Pre-requisites

Fundamental knowledge of applied mathematics, science and engineering

Type of Course

Practical

Course Domain

Core

-

Cognitive: Understand, Apply, Analyze, Evaluate, Create

Skills Imbibed

Affective: Awareness, Respond, Value, Organize

Psychomotor: Perception, Imitation, manipulation, articulation

Course Assessment Methods:

Practical Journal Assessment, Internal Oral Examination and External Practical Examination

Course Objectives:

To give the student

- 1. To give fundamental knowledge of fluid, its properties and behavior under various conditions of internal and external flows.
- 2. Understand the dynamics of fluid flows and the governing non-dimensional parameters,
- 3. Apply concepts of mass, momentum and energy conservation to flows,
- 4. Practice in the analytical formulation of fluid mechanics problems using Newton's Laws of motion and thermodynamics
- 5. To imbibe basic laws and equations used for analysis of static and dynamic fluids.
- 6. Apply fundamental principles of fluid mechanics for the solution of practical

Course Outcomes:

- 1. An understanding of fluid mechanics fundamentals, including concepts of mass and momentum conservation.
- 2. An ability to apply the Bernoulli equation to solve problems in fluid mechanics.
- 3. An ability to apply control volume analysis to problems in fluid mechanics.
- 4. An ability to use potential flow theory to solve problems in fluid mechanics.
- 5. An ability to perform dimensional analysis for problems in fluid mechanics.
- 6. A knowledge of laminar and turbulent boundary layer fundamentals.
- 7. An understanding of how fluid mechanics applies to biological, environmental, and micro-fluidic systems

Practical List

- *at List* 1. Venturimeter
- 2. Orifice meter
- 3. Reynolds's experiment

:

- 4. Bernoulli's experiment
- 5. Flow through helical coils
- 6. Flow through annular pipe
- 7. Flow through pipe & pipe fittings.
- 8. Flow through spiral coils.
- 9. Flow through packed bed
- 10. Flow through fluidized bed.
- 11. To study the properties of Newtonian and Non-Newtonian fluids.
- 12. Demonstration of -a) Rotameter b) Pitot tube
- 13. Two phase flow system

Lab Manual:

1. Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow.

Reference Books

REFERENCES:

- 1. Steeter U.L, 'Fluid Mechanics' V ed. Mc graw Hill Book Co., International Edn. 1971.
- 2. Richardson J.E. and Coulson J.M. Chemical Engineering. 3rd ed. Vol. 1 Pergamon Press 1985.
- 3. Miohell B.I. Fluid and Particle Mechanics Pergamon Press 1970.
- 4. Gupta S.K., Momemtum Transfer Operations, Tata McGraw Hill, 1979

Class & Semester	:	S. Y. B.Tech (Food Technology), Part II, Semester IV						
C Til	<u> </u>	Facing was a ball Charling Desirat World			110004			
Course Title	:	Environmental Studies Project Work	Course Code:	:	HS221			
Teaching Scheme (Hours)	:	2 hr /week= 2 x12= 24 hours	Credits	:	Nil			
Evaluation Scheme (Marks)	:	IPE : 30 EPE : Nil EOE : Nil	Duration of SEE for Theory part	:	2 hours (SEE at the yearend)			
Revision	:	Third	Month	•	December 2016			
Pre-requisites	:	Knowledge of fundamentals of Physics and Chemistry						
Type of Course	:	Filed work with necessary laboratory exp	Filed work with necessary laboratory experimentation					
Course Domain	:	Humanities and Applied Science	Humanities and Applied Science					
Skills Imbibed	:	,	Cognitive: Understand, Apply, Analyze, Evaluate, Create Psychomotor: Imitation, manipulation, articulation, naturalization					
Course Assessment Methods: Students Project/ field work assessment. However, their overall response during entire semester is also considered for evaluation.								
Practical List	ractical List :							
Field work under the supervision of course coordinator.								
Lab Manual :								
Institute's Laboratory Course Manual and equipment wise Standard Operating Procedure to follow in case of use of related apparatus, equipment.								

Class & Semester : S. Y. B.Tech (Food Technology) Part II, Semester IV										
Course Title	:	Introduction to	Per	formi	ng Arts			Course Code:	:	HS222
Teaching Scheme (Hours)	:	2 hr /week= 2 x1	2 hr /week= 2 x13= 26 hours					Credits	:	Nil
Evaluation Scheme (Marks)	:	Assignments Viva voce	:	50 25	Written Test Grand Total	:	25 100	Duration of Exam	:	Not Applicable
Revision:	:	Third						Month	:	December 2016

Pre-requisites	:	In order to conduct the course successfully, student's involvement and interest in the classroom is the pre- requisite.
Type of Course	:	Audit Course at institute level
Course Domain	:	Humanity and Fine Arts
Skills Imbibed	:	Cognitive: Understand, Apply Affective: Awareness, Respond, Value, Organize Psychomotor: Perceive, Imitate, Manipulate, Articulate, Adapt

Course Assessment Methods:

The students will be given five assignments each for 10 marks. At the end of the course, there will be a written test of 25 marks and a viva voce of 25 marks. There will be assessment for a total of 100 marks. Based on the marks obtained, they will be awarded with a grade similar to other credit courses. Though it is an audit course, obtaining passing grade is essential.

Course Objectives:

- 1. To understand the history of arts.
- 2. To cultivate and enhance the interest in Music and other performing arts.
- 3. To highlight that these arts are not only the medium of entertainment but also a medium for proper channelization of emotions as this plays a vital role in determining the quality of life.
- 4. To form and defend value judgments about music.
- 5. To acquire audience skills such as listening and viewing responsibly.
- 6. To understand & develop skills to become lifelong learners in the musical art, both as participants and as audience members.

Course Outcomes:

- 1. Students will be able to learn Fundamentals and types of Music and other allied arts.
- 2. Students will be able to analyze, appreciate, and interpret significant works of art.
- 3. Students will demonstrate critical thinking through analysis and evaluation of works of art.
- 4. Students will develop good listening and viewing skills.
- 5. Students will be able to understand the 'Gharana' system in Music.
- 6. Students will understand the classification of Musical instruments.
- 7. Students will demonstrate mastery of their designated area of concentration.
- 8. Students will demonstrate comprehension of global perspectives in visual culture.

Curriculum Content	Hours			
Unit I: Introduction to Music, Dance & Drama, History of Indian Music, Various Forms of Vocal Music.				
Unit II: History and introduction of Drama, Bharat muni natya shastra, street play, Sanskrit natya, Marathi	04			
sangit rangbhumi	04			
Unit III: Dance, its type, greek and roman theatres,	04			
Unit IV: Concept of Raga, Concept of Taal.				
Unit V: Notation System, Study of Gharana system in Music, Classification of Indian Instruments,				
Instrumental Music.	05			
Unit VI: Contribution of Great Musicians, Appreciation of Music. Performance of a Music Concert.				

- 1. 'Sangeet Visharad', Vasant, Sangeet Karyalaya, Hatras Prakashan.
- 2. Suchita Bidkar, 'Sangeet Shastra Vigyan', Sanskar Prakashan.
- 3. Sudhir Mainkar, 'Sangeet Kala Aani Shikshan', Sanskar Prakashan.
- 4. Bhaskar Chandavarkar, 'Vadyavedh', Sanskar Prakashan.
- 5. Arvind Mulgaonkar, 'Tabla', Popular Prakashan.
- 6. Chris Hogget ,'All about theatre-Off stage'.
- 7. Mrinalini Sarabhai, 'Understanding of Bharat Natyam'.
- 8. Joan Borysenko, 'Minding the body and mending the mind',.
- 9. V.K.Subbanna ,'Ragadalli Antrang'.

Equivalence of Second Year B.Tech (Food Technology) Semester III and IV

The above detailed syllabus is a revised version of the Second Year B.Tech(Food Technology)Program being conducted by the Shivaji University at the Technology Department of the University. This syllabus is to be implemented from June 2017, (Academic year 2017-18). The prime feature of this revision is the transformation of the existing curriculum into the Outcome based curriculum as specified in NBA rules and regulations.

The Equivalence for the subjects/courses of Food Technology at Second Year B Tech Semester III and IV pre-revised Program under the faculty of Engineering and Technology is as follows.

Second Year B. Tech Semester III (Food Technology)

Sr.No	Second YearB.Tech(Food Technology) Semester III Pre-revised syllabus	Second Year B.Tech(Food Technology) Semester III Revised syllabus	Remark
1.	Engineering Mathematics-III	Engineering Mathematics-III	Slight change in no of lectures and content
2.	Principles of Food Preservation	Principles of Food Preservation	Slight change in no of lectures and content
3.	Food Chemistry	Food Chemistry	Slight change in no of lectures and content
4.	Food Microbiology	Food Microbiology	Slight change in no of lectures and content
5.	Process Fluid Mechanics	Heat Transfer	Shifted to Semester IV with slight change in title, content and no of lectures
6.	Programming Practices	Programming Practices for Food Technologists	Modification in the content and title. Slight change in no of practicals and content
7.	Principles of Food Preservation Lab	Principles of Food Preservation Laboratory	Change in nomenclature. Slight change in no of practicals
8.	Food Chemistry lab	Food Chemistry laboratory	Change in nomenclature. Slight change in no of practicals
9.	Process Fluid Mechanics Lab	Heat Transfer Laboratory	Shifted to Semester IV with slight change in title. Slight change in no of practicals
10.	Food Microbiology lab	Food Microbiology Laboratory	Change in nomenclature. Slight change in no of practicals
11.	Introduction to Foreign Language	-	Shifted to semester V
12.	Environmental Studies	Environmental Studies	Change in no of lectures
13.		Soft Skills Development	Shifted from semester V with title change

Second Year B.Tech Semester IV (Food Technology)

Sr.No	Second YearB.Tech (Food Technology) Semester IV Pre-revised syllabus	Second Year B.Tech(Food Technology) Semester IV, Revised syllabus	Remark
1.	Food Process Engineering-I	Food Process Engineering-I	Slight change in no of lectures and content
2.	Food Additives and Contaminants	Food Additives and Contaminants	Slight change in no of lectures and content
3.	Food Biochemistry	Food Biochemistry	Slight change in no of lectures and content
4.	Fundamentals and Applications of Heat Transfer	Fluid Mechanics	Shifted to semester III with change in nomenclature, no of lectures and content
5.	Human Nutrition	Human Nutrition	Slight change in no of lectures and content
6.	Food Process Engineering-I lab	Food Process Engineering-I laboratory	Slight modification in nomenclature. Slight change in no of practicals and content
7.	Food Additives and Contaminants lab	Food Additives and Contaminants laboratory	Slight modification in nomenclature. Slight change in no of practicals and content
8.	Food Biochemistry lab	Food Biochemistry laboratory	Slight modification in nomenclature. Slight change in no of practicals and content
9.	Fundamentals and Applications of Heat Transfer lab	Fluid Mechanics laboratory	Shifted to semester III with change in nomenclature and no of practical
10.	Human Nutrition lab	Human Nutrition laboratory	Slight modification in nomenclature and no of practical
11.	Introduction to Performing Arts	Introduction to Performing Arts	Slight modification in the content

Audit course have not been assigned any credits. The students will be evaluated for these courses by the concerned course in charge. There will be grade conferred to the student. The grade will be based on conversion of marks obtained (Obtaining passing grade is essential). Please refer to chart in the detail examination scheme. The chart shows the marks range and the respective grade.